

Aya Al-majali

Sulfonamides antibacterial agents (The first synthetic antibacterial agents)

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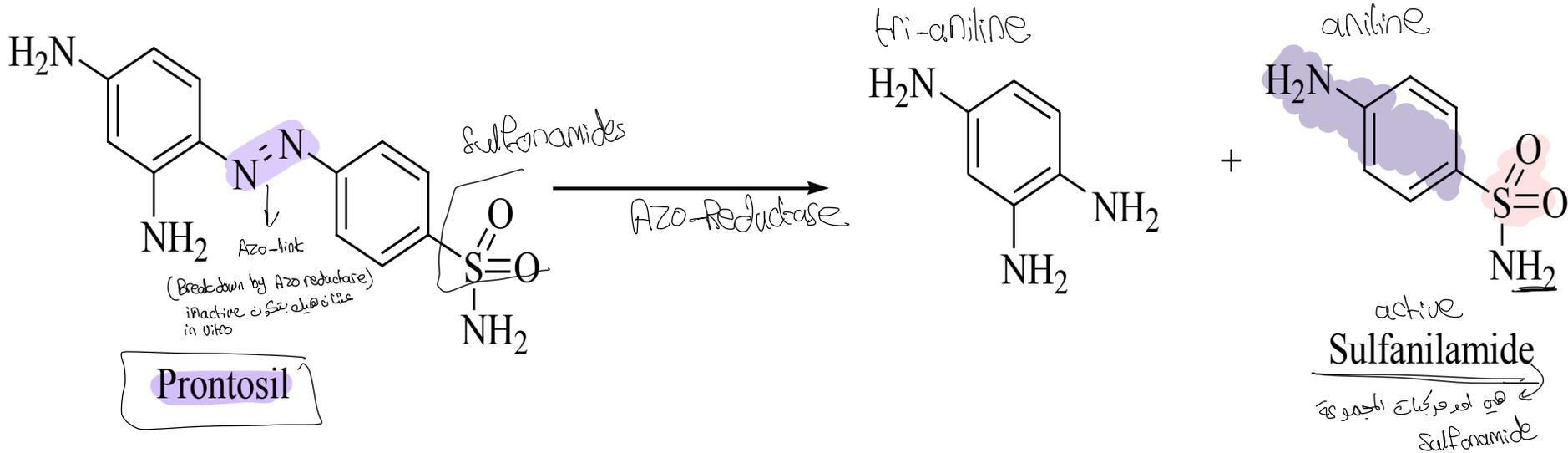
→ not antibiotics/antibiotics = anti cancer (natural)

Sulfonamides antibacterial agents

(Prontosil) بابوا ابرين مطبا على صيغة كبريتا اسول

Inactive
in vitro

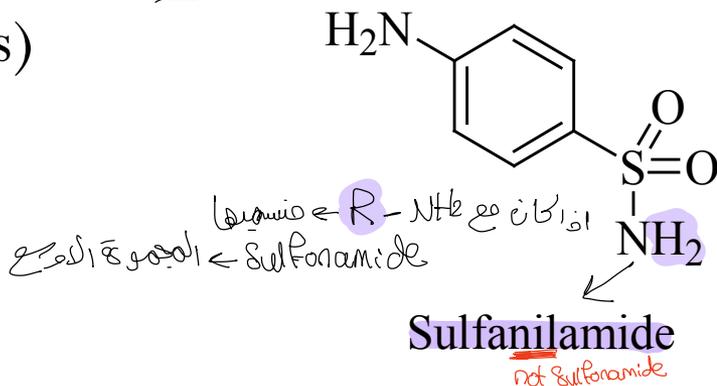
- In 1932. Domagk began to study a brilliant red dye called prontosil. This dye showed in-vivo antibacterial activity while it was in-vitro inactive (Prontosil is inactive on bacterial culture). active in vivo
- Later it was found that prontosil has to be activated by the in vivo metabolic pathways to give the active form.



والسبب انو هيا الصيغة هيا active فيه $\left(\times \right)$ و الـ 1000 انو هيا البكتريا تقدر Azo reductase طافه جسم البكتريا بـ عكس هيا و الـ 4
 Sulfasalazine $\left(\times \right)$ علاج هذا الـ 1000 تحت تـ كـ فـ

Nomenclature of Sulfonamides (Cont.)

- I. Antibacterials that are aniline-substituted . Sulfonamides (the Sulfanilamides)



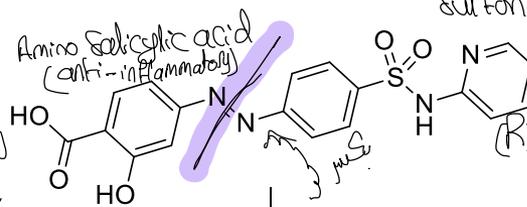
- II. Prodrugs that react to generate active Sulfanilamides (i.e

بـ علاج
 chronic
 bowel disease
 خصوصاً أما انه تخلي جسم البكتريا يعالج
 الجهاز الهضمي ويصل تقدمات فيها

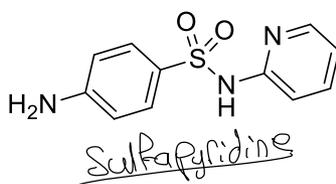
Sulfasalazine

anti-inflammatory
 Salicylic acid

anti-bacterial
 Sulfonamide



Azo reductase



$\left(\times \right)$ first line to treat IBD \rightarrow ciprofloxacin

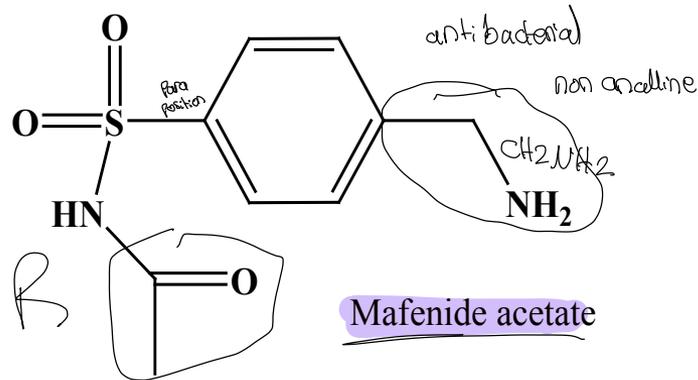
$\left(\times \right)$ second line IBD \rightarrow Sulfonamide

irritable bowel disease

Nomenclature of Sulfonamides

علاج
الحرارة

- III. Nonaniline sulfonamides (i.e., mafenide acetate)



Sulfonamides antibacterial agents

- Their bacterial activity is mainly on gram +ve and -ve bacteria
- limitation of the sulfa drugs use:

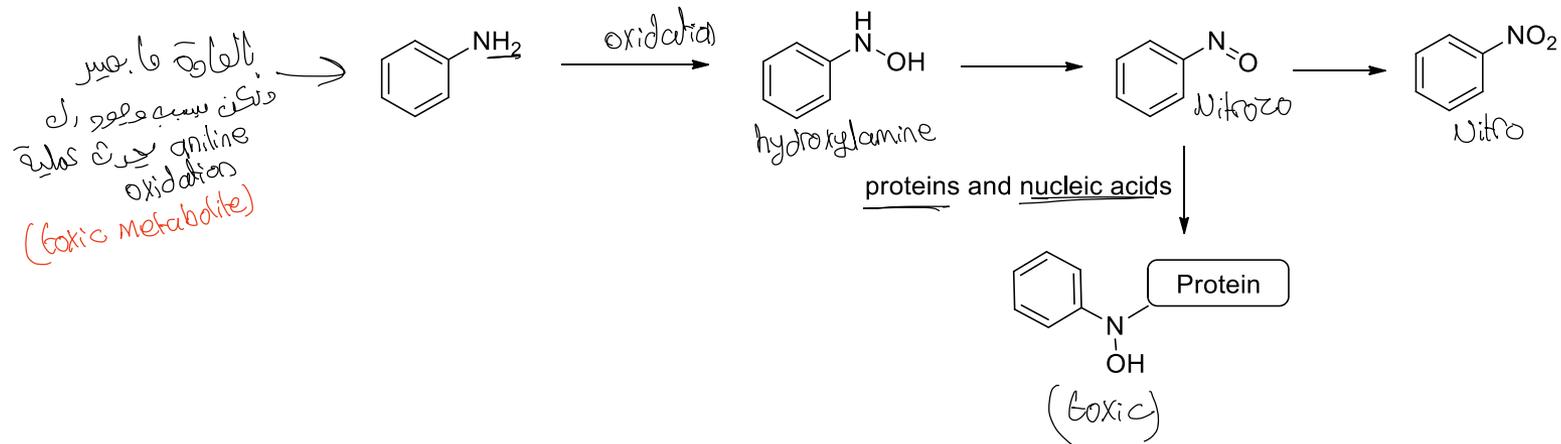
1 • Sulfa allergic reactions.

تكون
سريع
الماء

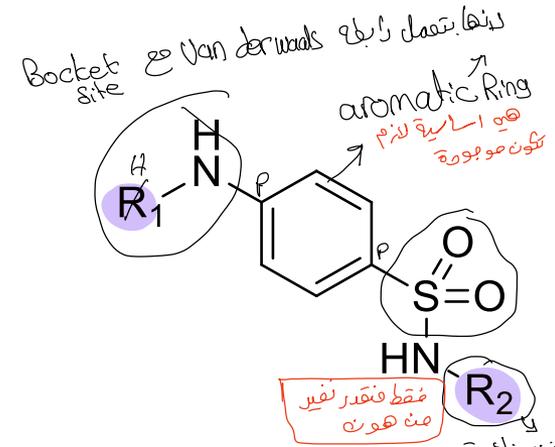
2 • The formation of crystalluria. ← ضمان S.A في الماء قليل خارج تعمل مشاكل في الماء منها

sulfonamide

3 • They give toxic metabolites after the oxidation of the aromatic amine:



SAR of sulfonamides



1. P-amino group is essential for activity and should be free (unsubstituted)

 - The sulfonamide nitrogen must have an attached hydrogen with a pKa similar to that of PABA (~6.5).
2. In the case of ^(Sulfasalazine) prodrugs the azo linkage that will be hydrolyzed to give the active free form.
3. The aromatic ring and the sulfonamide group are important for activity.
4. The sulfonamide and the amino group must be directly attached to the ring and in P position to each other.
5. Any extra substitution will reduce activity.
6. Sulfonamide nitrogen must be either ^{R-1H2} primary or ^{R-1HR} secondary.

لزيادة قابلية الذوبان الكهنة طفلة ان unme لتسهل ما يبعد Crystal area ← منط (electron withdrawn group) كته يفتقد H

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هم لتصنيع الحمض النووي لثديروفيثية وال DNA

folic acid is methyl donor

SA (46) يعتمد خلايا البكتريا على يؤثر على خلايا الانسان لديه

الانسان لا يتأثر بمنتجات التزيم Dihydropterolate Synthetase لكنه يتأثر المنتج النهائي له هو Folic من الطعام على العكس البكتريا مع يتأثر وتقل كمية Folic acid الذي يعتبر Methyl donor مع لتصنيع DNA الخلايا

Mechanism of action

- Sulfonamides are a competitive reversible inhibitors of dihydropteroate synthetase which is a vital enzyme for the synthesis of tetrahydrofolate (Coenzyme E).

هو يحدو تكثرة فتع البكتريا وغير لازم للانسان (هم لتصنيع الحمض النووي)

Tetrahydrofolate is important for pyrimidine nucleic acid synthesis so the bacteria can no longer grow and divide which gives time for the host immune system to destroy the bacterial cells.

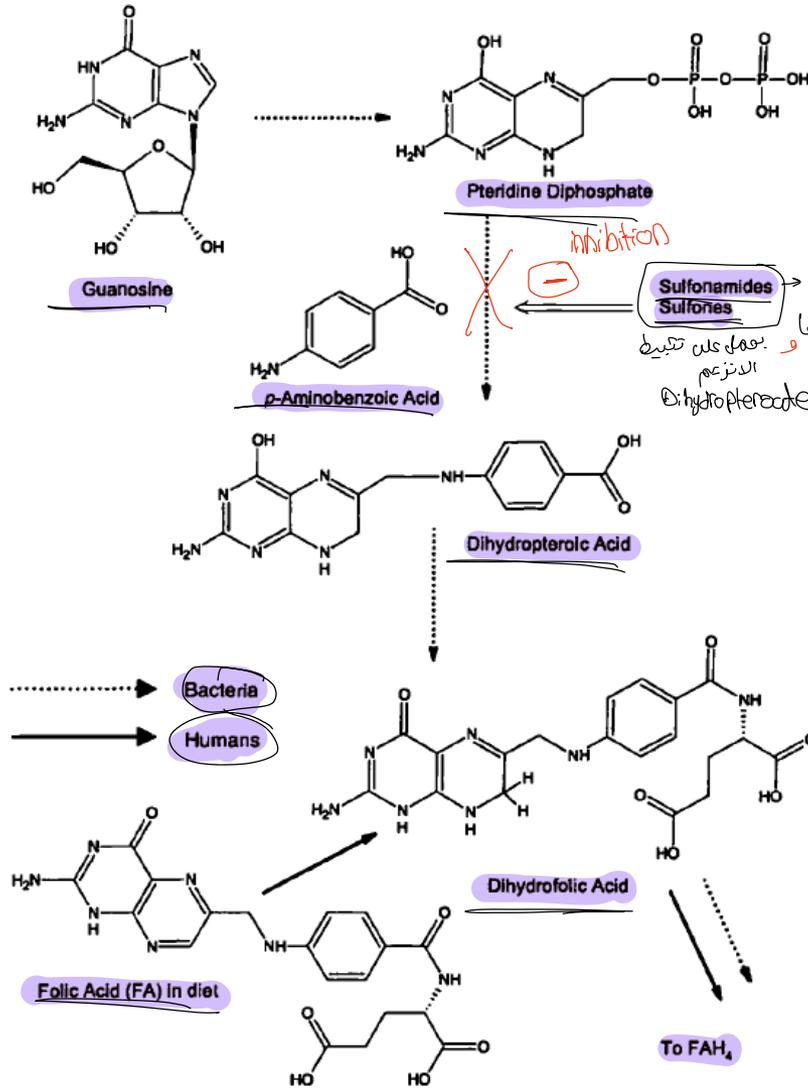
- Because of that sulfonamides have bacteriostatic effect not bactericidal so is not recommended in patients with weak or impaired immune system

لانه باثروا على التزيم

weak immune system static

Sulfonamide لا يضر البكتريا لكن لا يضر الانسان (Sulfonamide لا يضر البكتريا لكن لا يضر الانسان) AIDS & weak immune system (AIDS و ضعف المناعة) Penicilline هو bacteriocidal (Penicilline هو bacteriocidal)

Mechanism of action



طريقة تصنيع ال Folic acid داخل البكتريا
 P-aminobenzoic acid + Pteridine + Guanosine
 البكتريا بحاجة لـ Guanosine لتصنيع ال Folic acid

Reference: Wilson and Gisvold's Textbook of Organic

انزيم ال Dihydropteroyl synthetase

مركب بين ال PABA + Pteridine
 لتتصنع ال Dihydropteroyl acid

ولكن ال Sulfonamides تجرد ال P-aminobenzoic acid
 لـ تصنيع ال Dihydropteroyl acid
 ال Sulfonamides يوقف ال PABA

في ال انسان ال Folic acid يتكون من ال PABA
 ال Folic acid يتوقف عن تصنيع ال Folic acid

بعد ال Dihydropteroyl + Guanine
 ال Dihydropteroyl (2H) reduction ال Tetrahydropteroyl acid (4H)
 ال Folic acid (4H) (هو ال فعال)

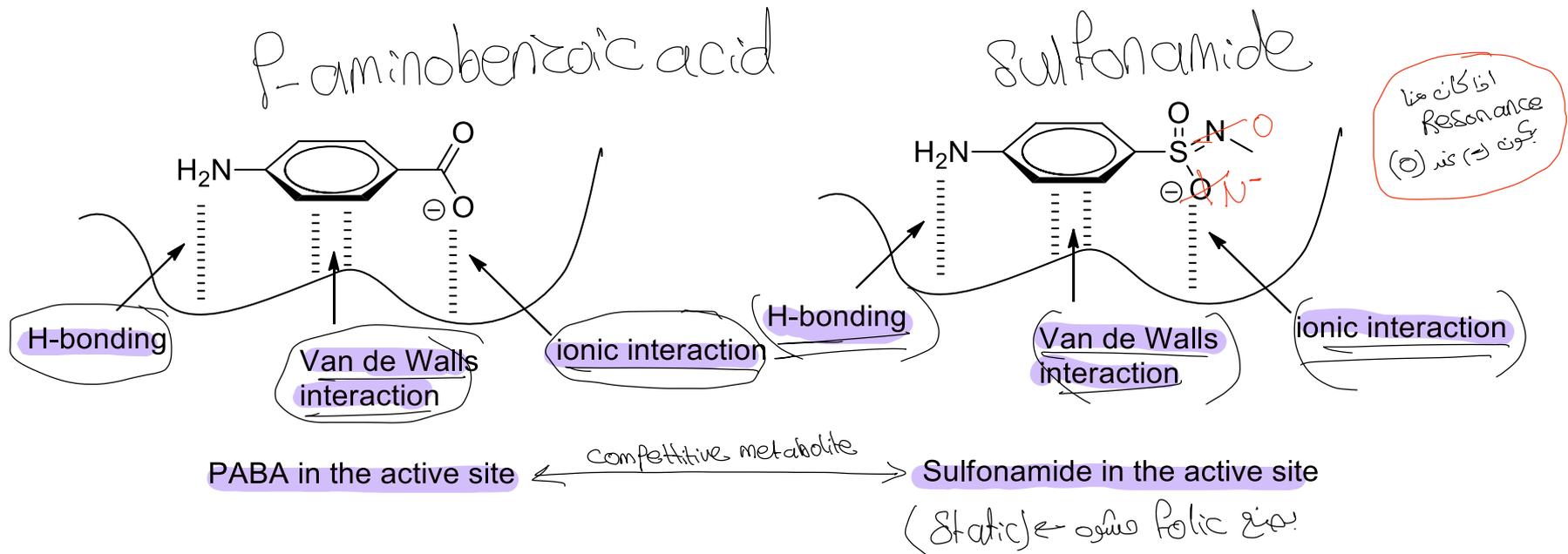
Figure 8-10 ■ Folate pathw in humans and bacteria and sites of inhibition by sulfonamides and trimethoprim.

Target of S.A → Dihydropteroyl synthetase

Mechanism of action

- Sulfonamides mimic *p*-aminobenzoic acid (PABA) which is the normal substrate for dihydropteroate synthetase. This means that sulfonamide will bind in the same manner as PABA:

يعني انه تركيزه اعلى هو الي يرتبط بال انزيم



Mechanism of action

- Because sulfonamides are reversible competitive inhibitors for the enzyme, the bacteria can increase the production of PABA to compete with sulfonamide at the active site and become resistant to sulfa drugs.
- In such case, the dose of sulfonamide agents should be increased to overcome this resistant mechanism. But this high dose is accompanied with an increase in side effects especially the crystalluria.
- N4 acetylation reduces drug solubility, which may result in precipitation in the urine leading to crystalluria. Increasing the pH of urine with a systemic alkalizer along with increased water intake will decrease the risk of this potential adverse effect.

Mechanism of action

- In human, the cell synthesized tetrahydrofolate from folic acid that obtained from food sources. This folic acid is normally transported to inside the cell by special transport system.
المصدر - الغذاء + Supplement
- Bacterial cell does not have such transport system and they should synthesize tetrahydrofolate using PABA.
- For that reason, human cells do not need dihydropteroate synthetase enzyme which means sulfonamides have selective antibacterial activity.

Mammalian Folate biosynthesis

⊛ Dihydrofolate Reductase
 يتم تثبيطه
 human: methotrexate
 Bacterial: trimethoprim
 Protozoal: pyrimethamine

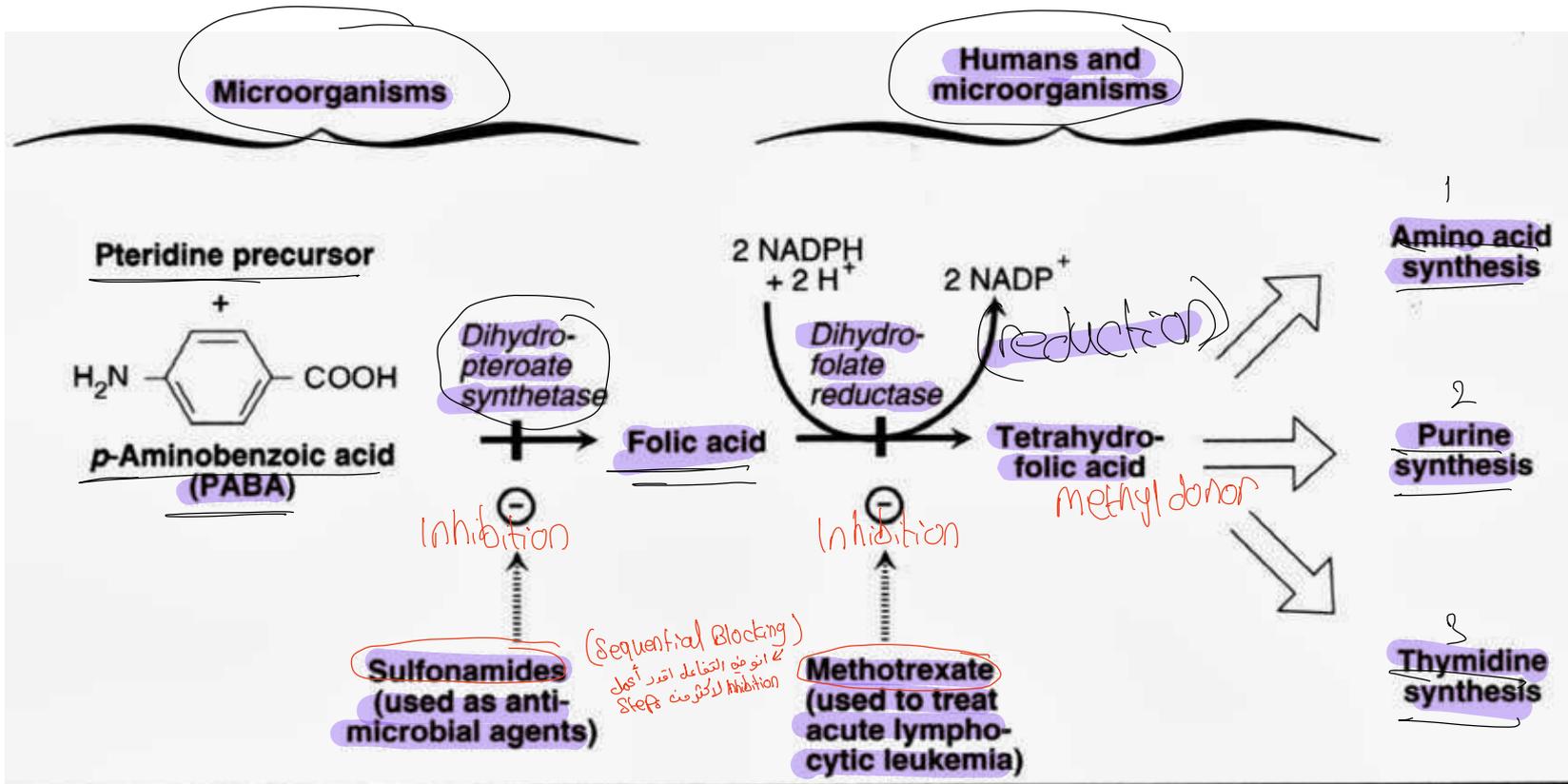


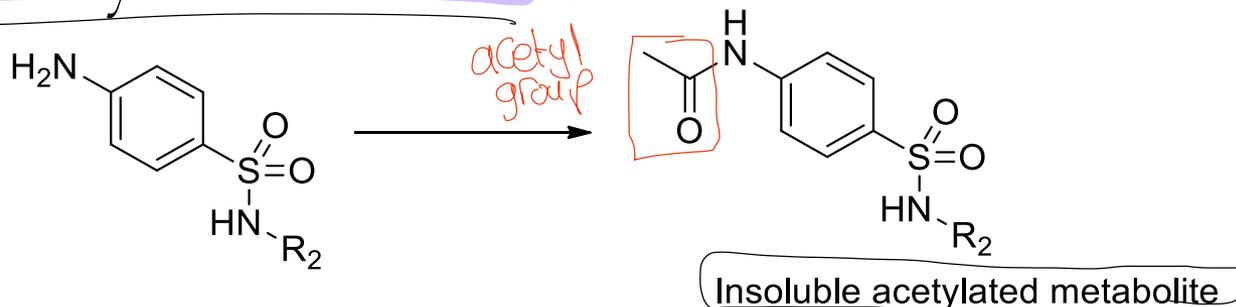
Figure 28.9

Inhibition of tetrahydrofolate synthesis by **sulfonamides** and **methotrexate**.

human

The problem of crystalluria

- Sulfonamides are mostly excreted in urine as acetylated metabolite.
- They are relatively water insoluble mainly due to the formation of the acetylated metabolites.



- The acetylated metabolite is non-ionizable under the pH conditions of the urine (≈ 7) that increase the possibility of precipitation and the formation of crystals in the urine (crystalluria).

The problem of crystalluria

- How to minimize the possibility of crystalluria formation with sulfonamides:

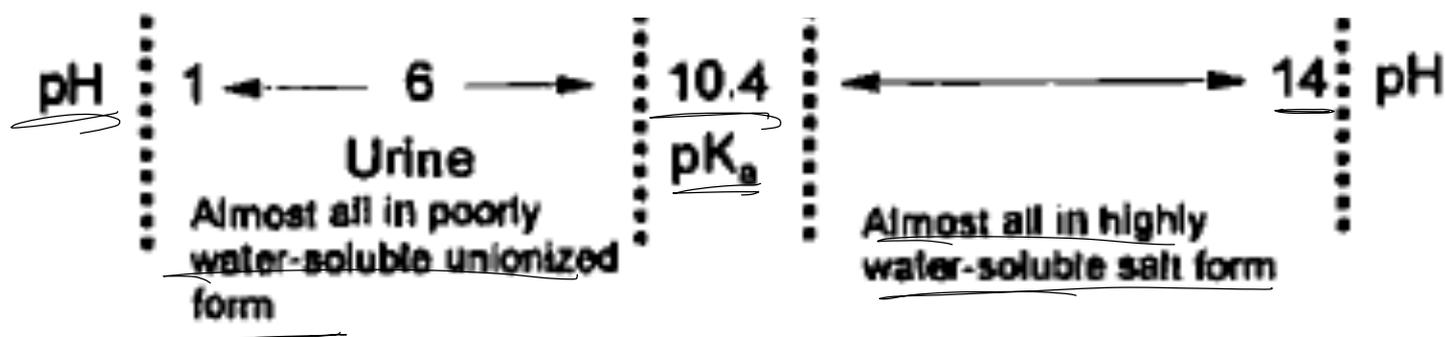
- (1) • Increase the urine flow. ↑
أقلية أكثر
مائية
- (2) • Increase the pH of the urine to increase the ionization of sulfonamides and the formation of water soluble salts (this can be done by taking sodium bicarbonate or potassium citrate. ← KOH
- (3) • Lowering the pKa of the sulfonamide group which will help to increase the ionization under the acidic conditions. This can be done by adding electron withdrawing group on the sulfonamide side chain ↓
مائي

أقلية أكثر
مائية →

TABLE 8-8 pK_a Values for Clinically Useful Sulfonamides

Sulfonamide	pK_a
<u>Sulfadiazine</u>	6.5
<u>Sulfamerazine</u>	7.1
<u>Sulfamethazine</u>	7.4
<u>Sulfisoxazole</u>	5.0
<u>Sulfamethoxazole</u>	6.1

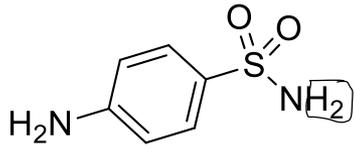
pKa > 6.5



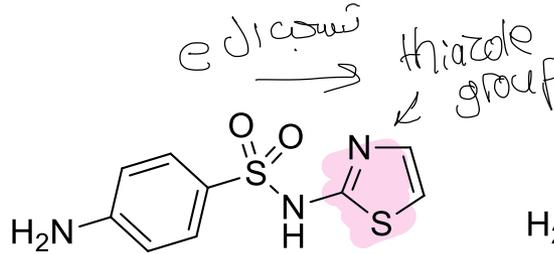
ع
طريقة
(م)

Sulfonamides with reduced crystalluria formation

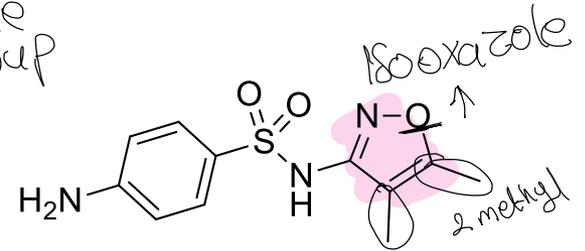
اذا كان ال pKa في وقتين لانه



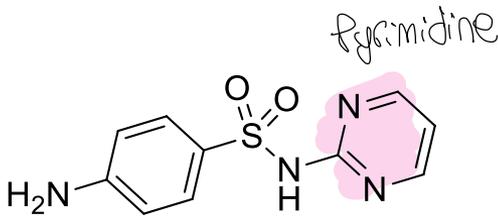
Sulfanilamide pKa = 10.4
ليس له تاثير



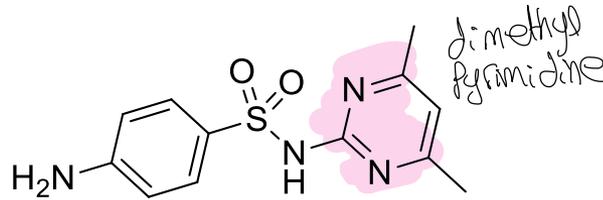
Sulfathiazole pKa = 8.5
اقل



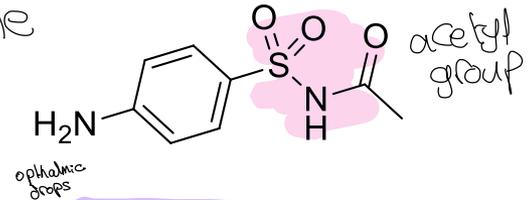
Sulfisoxazole pKa = 5.0
اكثر تاثيره باكثر تاثير



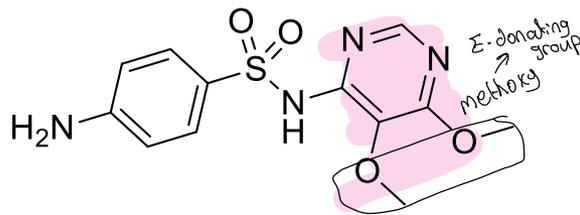
Sulfadiazine pKa = 6.5



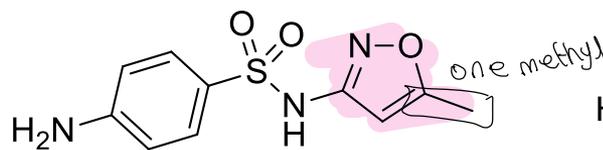
Sulfamethazine pKa = 7.4



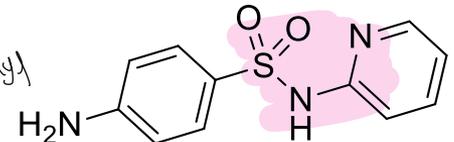
Sulfacetamide pKa = 5.4



Sulfadoxine pKa = 8.1



Sulfamethoxazole pKa = 6.1



Sulfapyridine pKa = 8.4

Clinical Application

The drugs are effective against both gram-positive and gram-negative organisms, but bacterial resistance and newer more effective drugs have replaced the majority of the previously available sulfonamides.

Today, many of the sulfonamides have been discontinued, but some are still available and are primarily limited to treatment of susceptible gram-negative organisms.

Products containing sulfonamides are shown in [Table 23.1](#).

Sulfisoxazole, in the form of the prodrug N1-acetylsulfisoxazole, is used in combination with erythromycin ethylsuccinate (EES) and indicated for the treatment of **otitis media**. → التهاب الأذن الوسطى

① **Sulfamethoxazole** in combination with trimethoprim (see below) is used to treat **uncomplicated urinary tract infections**,

② while sulfadiazine when combined with the antiprotozoal agent (pyrimethamine) is used to treat **Toxoplasma gondii infections**. → مرض الفتق
في كبد
الإنسان

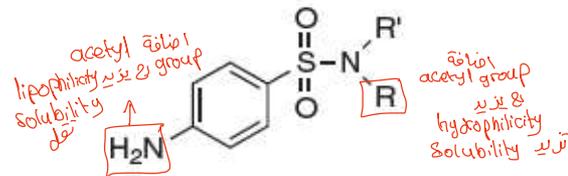
③ **Silver sulfadiazine** is used topically to treat burns, with both the sulfa drug and the silver ion having antibacterial activity.

④ **Sodium sulfacetamide** is a water-soluble preparation used to treat ophthalmic infections, while **sulfasalazine** is effective in the treatment of **ulcerative colitis**.

It is only poorly absorbed from the GI tract where it is hydrolyzed by intestinal

Clinically relevant sulfonamides

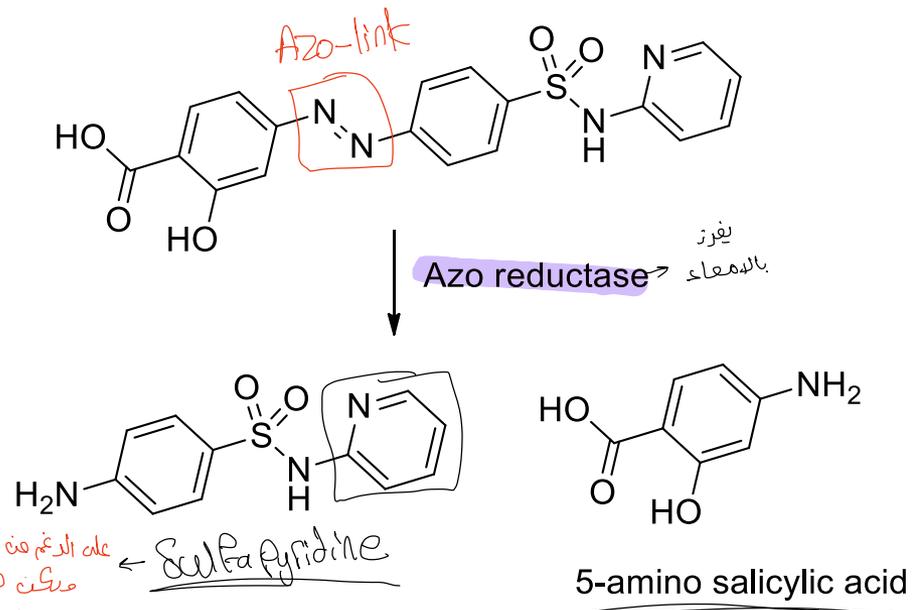
Clinically Relevant Sulfonamides



Drug: Generic Name	Product	R	R'	pKa
<u>Sulfisoxazole acetyl (prodrug)</u> <i>more hydrophobic</i>	In combination with erythromycin ethylsuccinate <i>otitis media (more hydrophobic) حتى لو ان acetylation</i>			5.6 after hydrolysis
<u>Sulfamethoxazole</u> (<u>Co-trimoxazole</u>)	In combination with trimethoprim		-H	5.0
<u>Sulfadiazine</u>	<u>Oral dosage form</u>		-H	6.52
<u>Silver sulfadiazine</u>	<u>Topical dosage form</u>		\ominus Ag \oplus	
<u>Sulfacetamide sodium</u> (<u>prodrug</u>) <i>more hydrophilic (قطرات العين)</i>	<u>Ophthalmic dosage form</u> <i>very soluble لا يتكون إعتاد imitation</i>		\ominus Na \oplus	5.4 free acid
<u>Sulfasalazine</u> (<u>prodrug</u>) <i>anti-bacterial anti-inflammatory</i>	<u>Gastrointestinal oral dosage form</u>			

Sulfonamide prodrugs

- **Sulfasalazine:**
 - Used in local intestinal infections.
 - Gives sulfapyridine and 5-aminosalicylic acid upon the breakdown of the azo bond.
 - Used mainly in ulcerative colitis.



على الرغم من أن $pKa=8.14$ عالية ومنه صامرينا للبرم العمل عليه لحتى يتخلص من (crystaluria) يمكن هذا الدواء اننا بديه اياه يكون هيكه مما انه طاعى بحد له امتصاص فيه المعدة والدمعاء

Other folate reductase inhibitors

- **Trimethoprim:**

- Inhibits dihydrofolate reductase: this enzyme has human homologue but they do not have that much similarity in structure.... Therefore trimethoprim is 1000 more active on the bacterial copy of this enzyme.

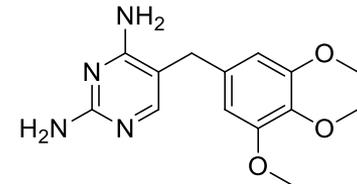
حلاينه ما باش كل
human
في فعاله انتر جانبيه

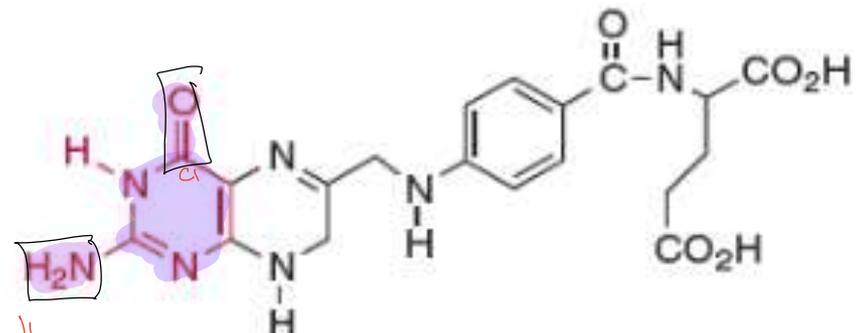
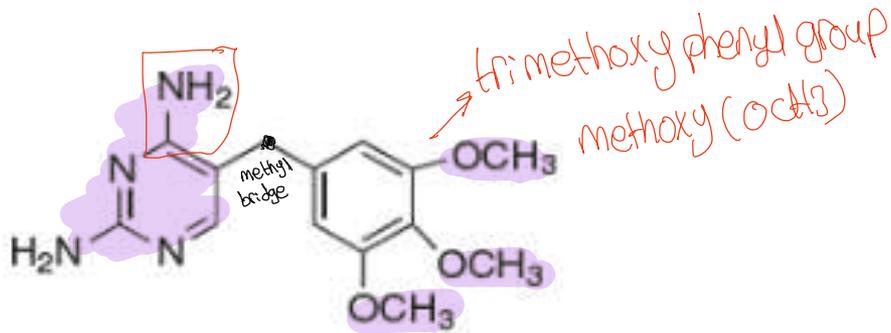
- Normally used in combination with sulfamethoxazole

(cotrimoxazole): كتر تجربه

- 1 • Lower dose from both drugs means less side effects. ^{→ crystalurea}
- 2 • More effective than the monotherapy since they are targeting two different enzymes in the same metabolic pathway... this is what is called sequential blocking.

Methotrexate → يمنع كاتر خلايا
سرطانية و لروما تيزوم
كفعل كل
الانسان
(تثبيته
inhibition
of human)





Trimethoprim (Proloprim, Trimpex)

Trimethoprim + sulfisoxazole (Co-Trimoxazole)

Trimethoprim + sulfamethoxazole (Bactrim, Septra)

المفروضة
(Trimethoprim + Sulfamethoxazole)
هو الاسم التجاري غالباً على الاسم العلمي
(معرفة الحق)

Dihydrofolic acid

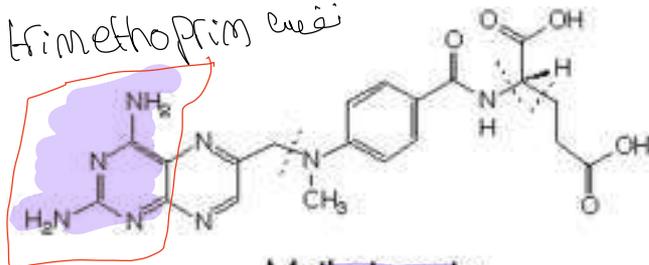
is natural
substrate of
dihydrofolate Reductase

(التي تالون لبعض الناس)

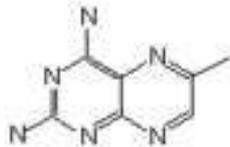
⊗ all Dihydrofolic acid + trimethoprim + methotrexate have the same pharmacophore units
is Pyrimidine + NH₂, but different in C1 → Dihydrofolic → C₁ is carbonyl &
trimethoprim → C₁ is NH₂
methotrexate → C₁ is NH₂
Binding pocket: (Reductase enzyme)

- There is another dihydrofolate reductase
- inhibitor –for Human enzymes (**Methotrexate**)
- which is used as anticancer drug . *+ Rheumatoid Reductase*
- There is another one too, (**pyrimethamine**), as a malarial
dihydrofolate reductase inhibitor

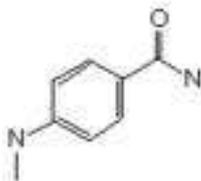
Pharmacophore trimethoprim esei



Methotrexate



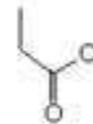
fragment1



fragment2



fragment3



fragment4

الكفاءة
البيولوجية

as combination

• Sulfonamide + Trimethoprim are :

(1) • Septra®

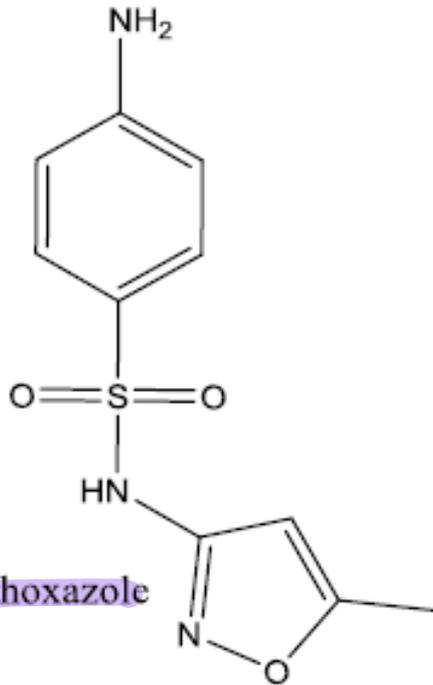
(2) • Balkatrin® . *treat gram (-)/gram (+) bacteria*

• Another use of Sulfadrug is for Protozoa (eukaryotic cells) like Amoeba & Malaria, and the most important use is to treat Pneumocystis carinii (Pneumonia) in AIDS patients, (Sulfamethoxazole + Trimethoprim).

Combination is static action

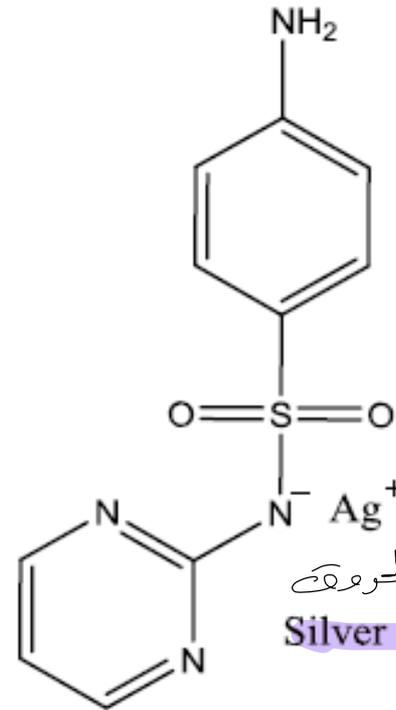
*Sulfadugs are divided to : Short (1) (Sulfacetamide), Moderate (2), Long acting (3).

UP



Sulfamethoxazole

DOWN



topical
للرقيقة
Silver Sulfadiazine

Protein binding of sulfonamides

- Vary in plasma protein binding: Sulfaisoxazole... ~~76%~~^{37%}, Sulfamethoxazole... 60%, sulfadiazine.... 38%.
- The fraction that is protein bound is not available for enzyme inhibition, therefore this fraction is inactive.
- The protein binding is a reversible process, so there will be a gradual release of sulfonamide which will become available.
- Factors affecting protein binding of sulfonamides:
 - (1) ● Lipophilicity of the structure.
 - (2) ● Substitution on the free amine will increase protein binding (such as the acetylated metabolite is more protein bound than the parent sulfonamide).

عند حفظ
التركاز

تقلل من فعالية الدواء
→ يزيد من half-life
العلاج

Use :

(balkutin) cleo

- The original Sulfanilamide was used against most of the infections, Upper and Lower respiratory tract infections (Pneumonia mainly), it was the only treatment available. ¹
- Their use was mainly for UTI (because it's eliminated quickly & their spectrum covers G-ve bacteria), it's still used until now. ² *first line → Cipro* *E. coli*
- Spectrum ³
- Broad spectrum (G+ve & G-ve) bacteria, with time, development of resistant happens ⁴

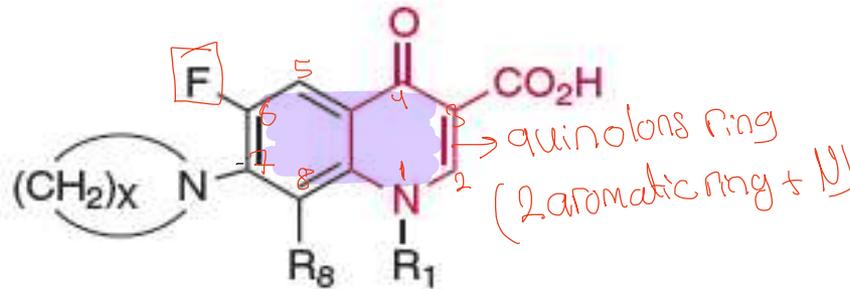
Protein binding of sulfonamides

لا بد من
مركبات
S.A

- Since albumin is basic, acidic and neutral drugs will primarily bind to albumin.
- If albumin becomes saturated, then these drugs will bind to lipoprotein.
- Basic drugs will bind to the acidic alpha-1 acid glycoprotein.
- Protein binding can influence the drug's biological half-life in the body but this relationship still not clear since some drugs with low protein binding have long duration of action (sulfisoxazole: protein binding 37% and half life is 17 hours).

acidic → albumin
basic → α-acid glycoprotein

(2) **4-Quinolones** (Synthetics) not antibiotics



4-Quinolone
(pharmacophore shown in red)

The **fluoroquinolones** have been found to be effective in treatment of various bacterial infections depending on the nature of the substitution on the 4-quinolone pharmacophore. (urinary tract infection)

MOA

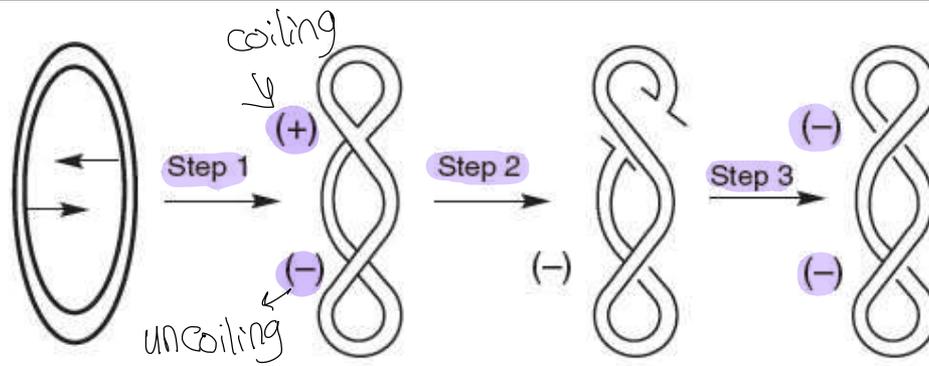
Irreversible inhibitors of DNA gyrase and topoisomerase IV, key enzymes involved in DNA-dependent RNA polymerase (DDRP).

⊕ إذا صنعت تصنيع
Protein في بروتين
Bacteriocidal
not static
Fluoroquinolones في بروتين
ما لا يترك على (DDRP enzyme)
بأنفسه على انزيمات هما
(gyrase + topoisomerase hormone)

هو اليه يصنع نسخ RNA

(DDRP)

الانزيم والمضغ
DNA ل
= RNA
* فانك هوية انزيم RNA المعتمد على
DNA

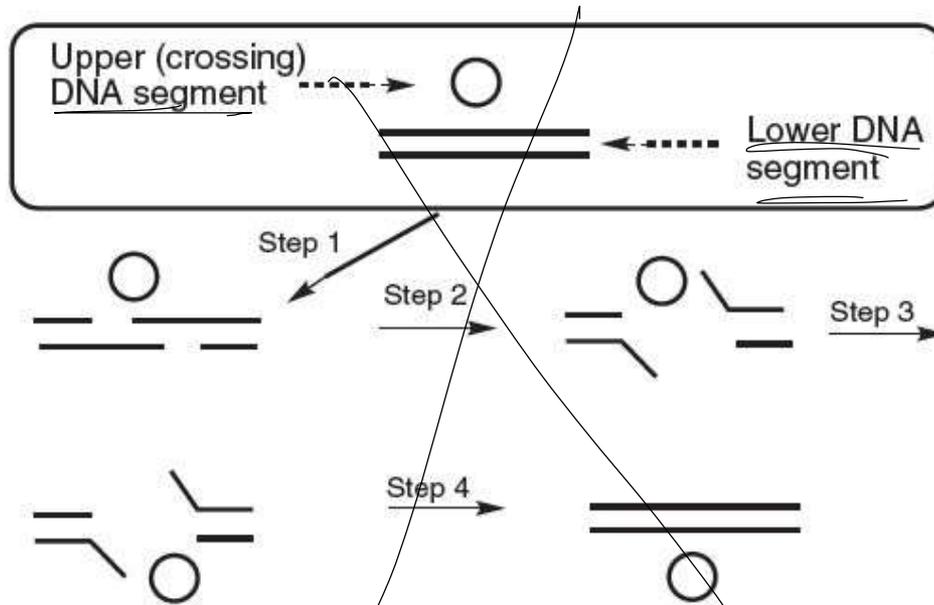


⊕ DNA إلى جسم الإنسان هي سلسلة طويلة كد شكل supercoiled (مفتولة) ولحقتها يشترك RNA Polymerase للبرم تبطل مشدودة والانزيمات (topoisomerase + gyrase) هما التي يفتحها الفتلة تبقت DNA
 ⊕ الفتلة عبارة عن (-) و(+) يعني وصية امام وثانية خلفه اما اذا كانت الفتلة (-) مرفوعه أو (+) نازل (مع فتلة الفتلة)
 ⊕ الية عمل الانزيم (topoisomerase + gyrase) -8

- (1) Stabilization of DNA
- (2) Break both strand of back segment + pass unbroken segment through the break and resealed on the front side

عملية التي
 gyrase +
 topoisomerase

A. View from the top: Step 1. Stabilize positive node. Step 2. Break both strands of the back segment. Step 3. Pass unbroken segment through the break and resealed on the front side.



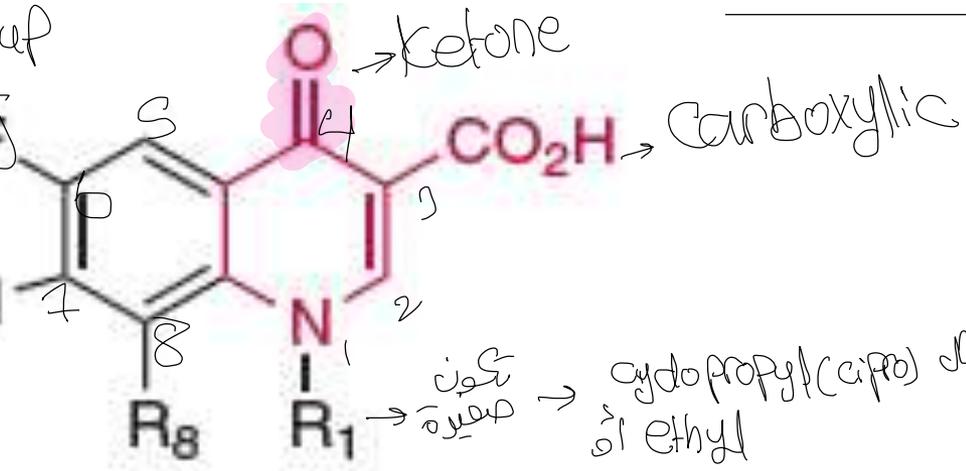
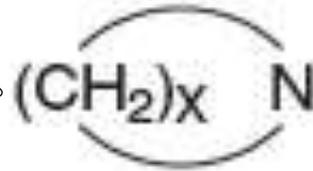
B. View from the side: Step 1. Staggered cuts in each strand. Step 2. Gate opens. Step 3. Transverse segment passed through the break. Step 4. Reseal cut segment.

يعني يقطعها ويقدمها للامام
 مبلزتها مع تحول الفتلة (-) و(+)
 مهيكل مع يشترك الانزيم

⊕ all halogen are hydrophobic group

⊕ هالوجينات لها تأثير توازن بين الـ hydrophilic + hydrophobic

Bi-Pyridine الحلقة →



تكون صغيرة → ethyl or cyclopropyl (cipro)

SAR

• The quinolone pharmacophore is essential for activity through binding to the DNA gyrase (Table 23.2).

• R_1 is important for potency and commonly consists of an ethyl or cyclopropyl.

Small size + hydrophobic → Fluoro at C6 improves penetration of the bacterial cell wall through improved hydrophobicity.

• Heterocyclic substitution at C7 affects the spectrum of activity against gram-negative bacteria.

• R_8 affects spectrum of activity as does R_1/R_8 linked forming a third ring in the molecule (flavoxacin).

(مفكرة ميون) of flavoxacin

الرباعي له structure تبعتها لانيد (SS)

Physicochemical and Pharmacokinetic Properties

ما يمنع الامتصاص
يشترك عليه مع
Ciprofloxacin
مع يتحول
(inactive)

- 4-Quinolones are incompatible with heavy metals (e.g., Ca²⁺, Mg²⁺, Zn²⁺, Fe²⁺, Al³⁺) due to an insoluble chelate resulting from bonding between the metal and the C3 carboxyl and C4 ketone.
→ تمنع امتصاص الطرفية
- 4-Quinolones may cause skin phototoxicity upon exposure to the sun (UV A radiation)
Side effect أوه



ADVERSE EFFECTS

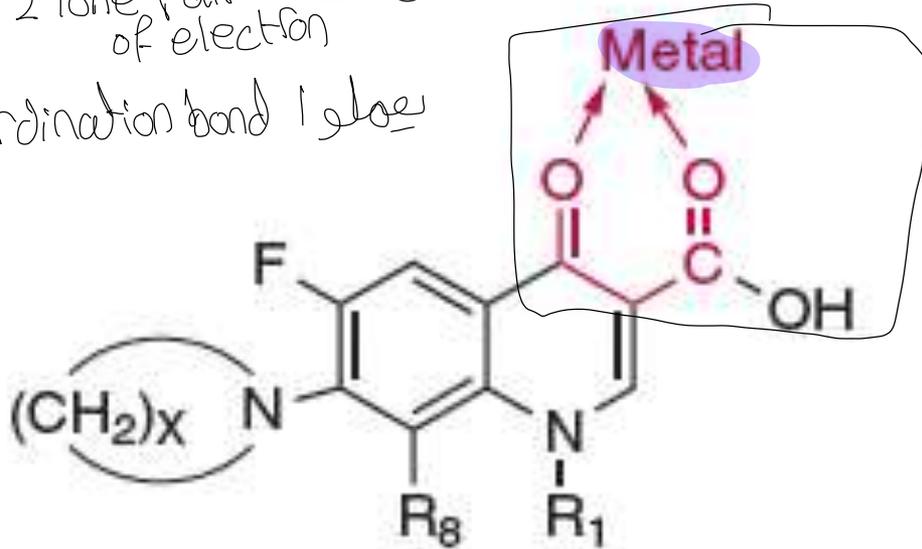
4-Quinolones: GI disturbance: nausea, vomiting, and abdominal discomfort. CNS effects: headache and dizziness, but may also include hallucinations, insomnia, and visual disturbances due to binding of lipophilic drugs to GABA receptors. Several analogs caused QT prolongation leading to their removal from the market.

⊗ مثالاً انزيمات الاله بتتسبب قطع حائل gyrase بتكونه (metaloenzyme) يمينه تحتوي على Metal من تركيبها الكيميائي
 وال Metal تعمل ← coordination bond وهي رابطة قوية جداً يصعب فكها

Binding of the drug to DNA gyrase involves the carboxyl and the ketone.

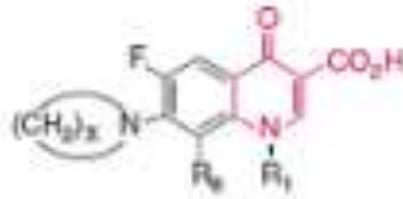
⊗ MOA of metalloenzyme of gyrase

2 Metal مع 2 lone pair of electron
 coordination bond
 @ carbonyl group (C3+C4)

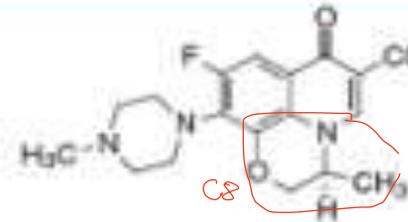


Metal chelation

مطلوبه فقط الي هكده ما الباعه
 لتعرفه المتنوع الي على structure



Drug: Generic Name	Trade Name	R ₁	N (CH ₂) _x	R ₆
Norfloxacin	Noroxin	C ₂ H ₅		H
<u>Ciprofloxacin</u>	<u>Cipro</u>			H
Gatifloxacin	Tequin			CH ₃ -O
Moxifloxacin	Avelox			CH ₃ -O
Gemifloxacin	Factive			
Besifloxacin	Besivance			Cl
Fluoroxacin*	Xtoro			N=C



Ciprofloxacin (Racemic) (Floxin)
 Levofloxacin (1-S) (Levaquin)

Clinical Application

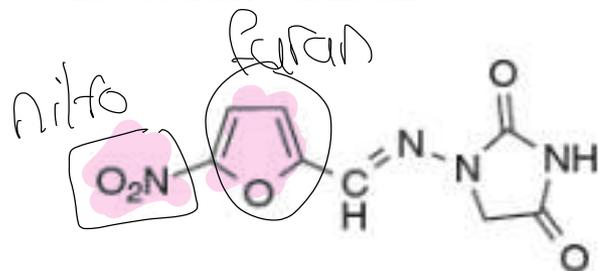
- The fluoroquinolones represent a potent class of bactericidal agents with utility in a variety of infectious conditions. The most common indications include UTIs caused by sensitive organisms; prostatitis; some sexually transmitted infections; respiratory infections; and bone, joint, and soft tissue infections.

bone + joint +
soft tissue
هذه ترميم عظمية
يعني ما بوسطهم الدواء
في انا بحاجة لعاء يكون
ال Volume of
Distribution عالي

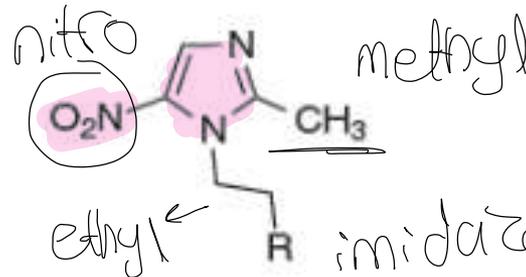
Synthetic

Nitroheteroaromatic Compounds

(3) **Nitrofurantoin** Furan (O + 2 other atoms) + Nitro



Nitrofurantoin
(Furadantin, Macrochantin)



Metronidazole, R = OH (**Flagyl**) anaerobic gram⁻ Bacteria
Tinidazole R = SO₂C₂H₅ (Fasigyn)
ethyl sulfonide

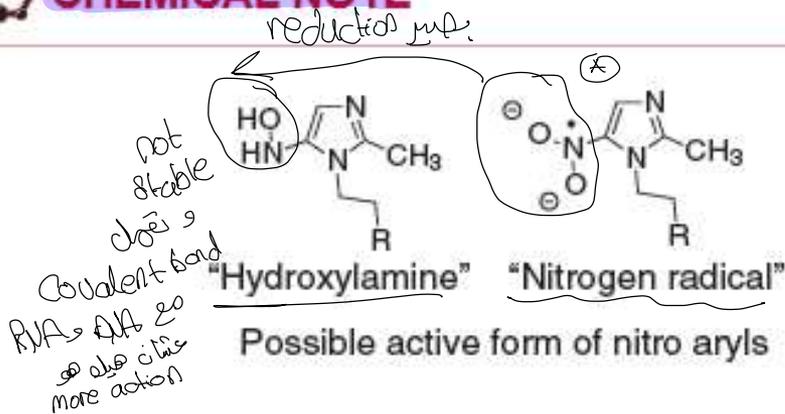
- **Nitrofurantoin** is the **only** nitrofuran which remains available and is used for treatment of uncomplicated **UTIs**.

Metronidazole

- Metronidazole and tinidazole are used to treat some bacterial infections (e.g., GI tract peptic ulcer, pseudomembranous colitis) and protozoal infections (e.g., giardiasis, trichomoniasis).



CHEMICAL NOTE

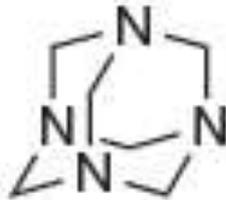


MOA ← MOA
موت الخلايا

- Most likely, the nitroheteroaromatic compounds are prodrugs in which the nitro group is reduced to the active hydroxylamine or nitrogen radical which interferes with DNA and or RNA.

Methenamine and Phosphomycin

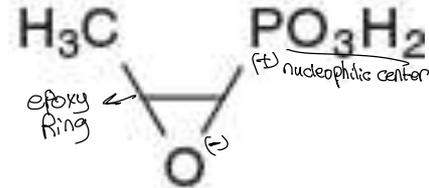
4N
6C



antibacterial

Methenamine

(Prosed, Urimax, Urised, Uroqid-Acid)



Phosphomycin

(Monurol)

limited values and are used in uncomplicated

UTIs

MOA

(* inactive in vitro)
* active in vivo

• **Methenamine** is a prodrug, which in acidic urine generates ammonia and formaldehyde. The latter forms a Schiff's base with bacterial protein resulting in antibacterial action.

مخبر
→ give → ammonia + formaldehyde

Bactericidal

• **Phosphomycin**, through alkylation of a key sulfhydryl group in a bacterial transferase essential in cell wall glycoprotein synthesis, inhibits bacterial growth.

→ alkylating agent →

also via
covalent bond
side chain of
Bacteria