



تفريغ كايبتك

Ch 3

محاضرة:

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الصيدلانية:



لجان الرّفعات

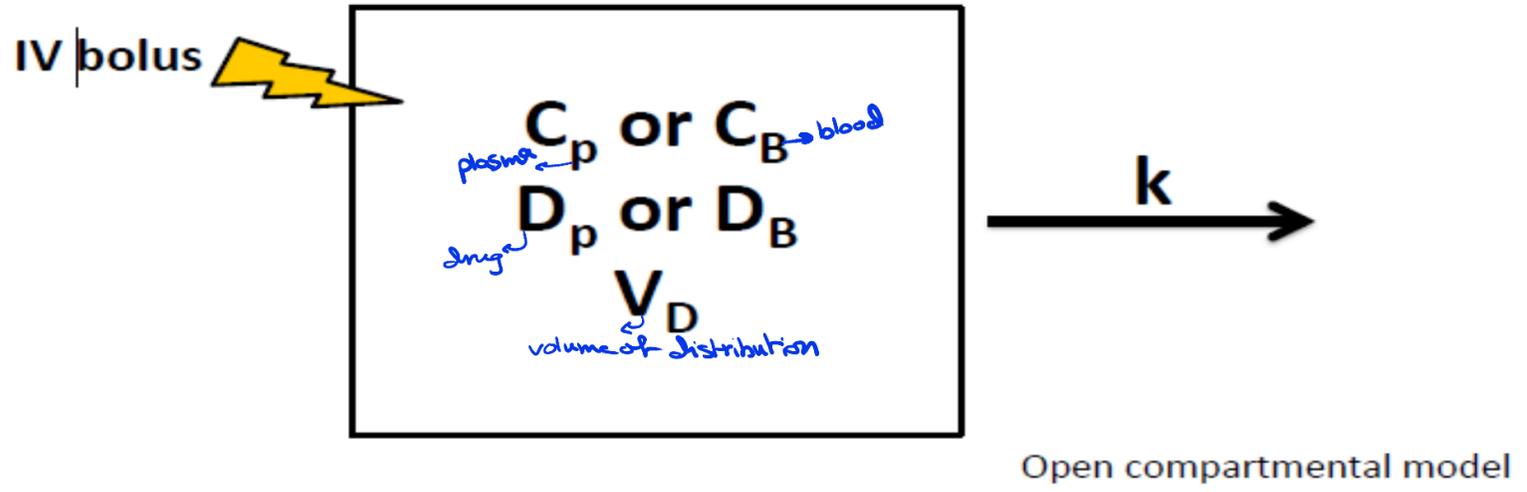


One compartment open model: intravenous bolus administration

PK theory lec.3

One-compartment model

- IV bolus- One compartment model:



Cont,

- The simplest kinetic model that describes drug disposition in the body is to consider that the drug is injected all at once into a box, or compartment, and that the drug distributes instantaneously and homogeneously (kinetically) throughout the compartment.
- Drug elimination also occurs from the compartment immediately after injection.

One-compartment model assumptions

Assumptions

- Rapid Mixing

drug is mixed instantaneously in blood or plasma.

- One compartment

Drug in the blood (plasma) is in rapid equilibrium with drug in the extravascular tissues.

- Linear Model

Drug elimination follows first order kinetics.

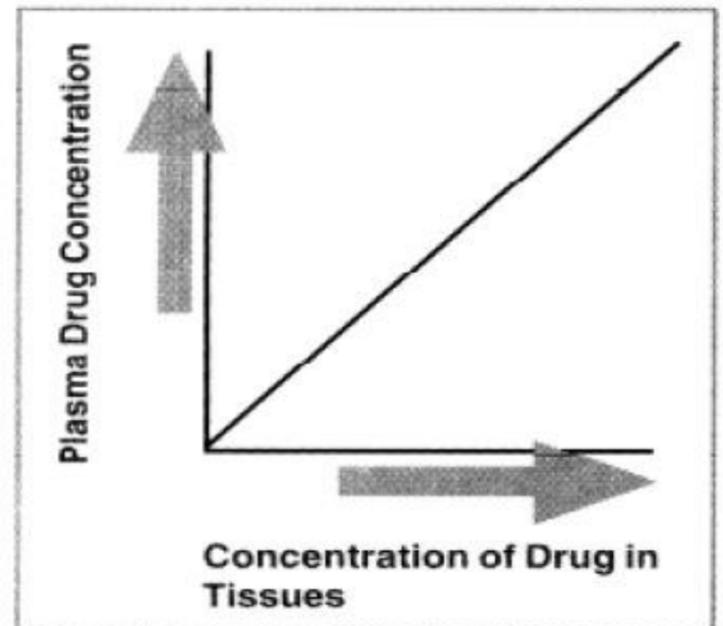


Figure 1.2. Relationship of plasma to tissue drug concentrations.

Cont,

- Changes in the plasma drug concentration reflect changes in drug concentrations in other tissues.

but proportional

- However, the plasma drug concentration does not equal the concentration at other sites but rather indicates how it changes with time.

- Generally, if the plasma concentration of a drug is decreasing, the concentration in tissues will also decrease.

Figure 1.3 is a simplified plot of the drug concentration versus time profile following an intravenous drug dose and illustrates the property of kinetic homogeneity.

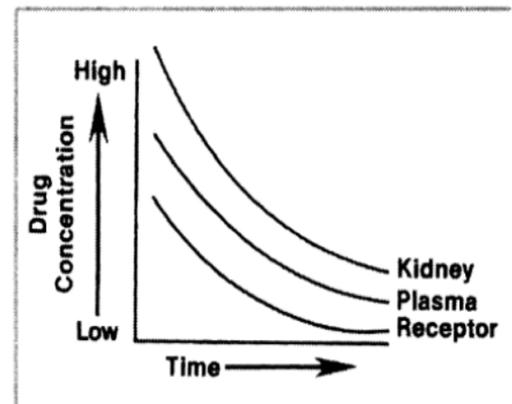


Figure 1.3. Drug concentration versus time.

Cont,

رابطته مع جميع الادوية

- The property of kinetic homogeneity is important for the assumptions made in clinical pharmacokinetics.
- It is the foundation on which all therapeutic and toxic plasma drug concentrations are established.
- That is, when studying concentrations of a drug in plasma, we assume that these plasma concentrations directly relate to concentrations in tissues where the disease process is to be modified by the drug (e.g., the central nervous system in Parkinson's disease or bone in osteomyelitis).
- This assumption, however, may not be true for all drugs.

Linear Kinetics (First order)

Elimination rate or change in concentration is proportional to the amount available for elimination.

Glomerular Filtration } Passive

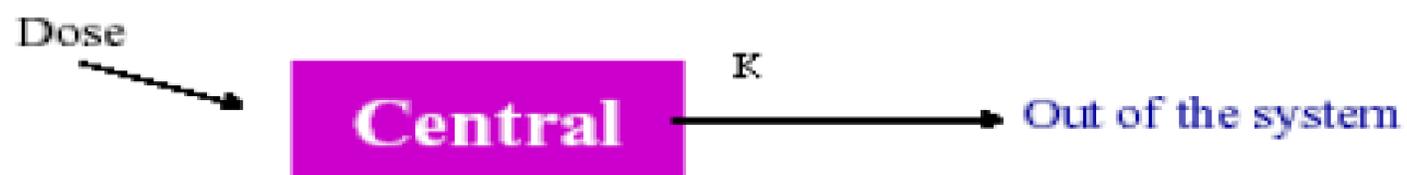
-Tubular secretion

-Biliary secretion } Involve enzymatic processes (active)

-Biotransformation

Cont,

1-Comp. Model: IV Bolus Dosing



X_t : the amount of drug remained in the compartment

K : first-order elimination rate constant (**OVERALL**)
(unit = time^{-1})

$$\text{Rate of elimination} = \frac{dX}{dt} = -KX$$

Elimination rate

Elimination rate is a first-order process



The elimination is dependent on the drug concentration or amount in the body

Elimination rate constant (k):

- 1st order rate constant
- Unit: 1/time (time⁻¹)
- $k =$ for all elimination processes = $k_e + k_m$ (Mainly)

$$k = k_e + k_m$$

k_e : 1st order rate constant of excretion

k_m : 1st order rate constant of metabolism

Cont,

- The rate of elimination from the compartment can be calculated as

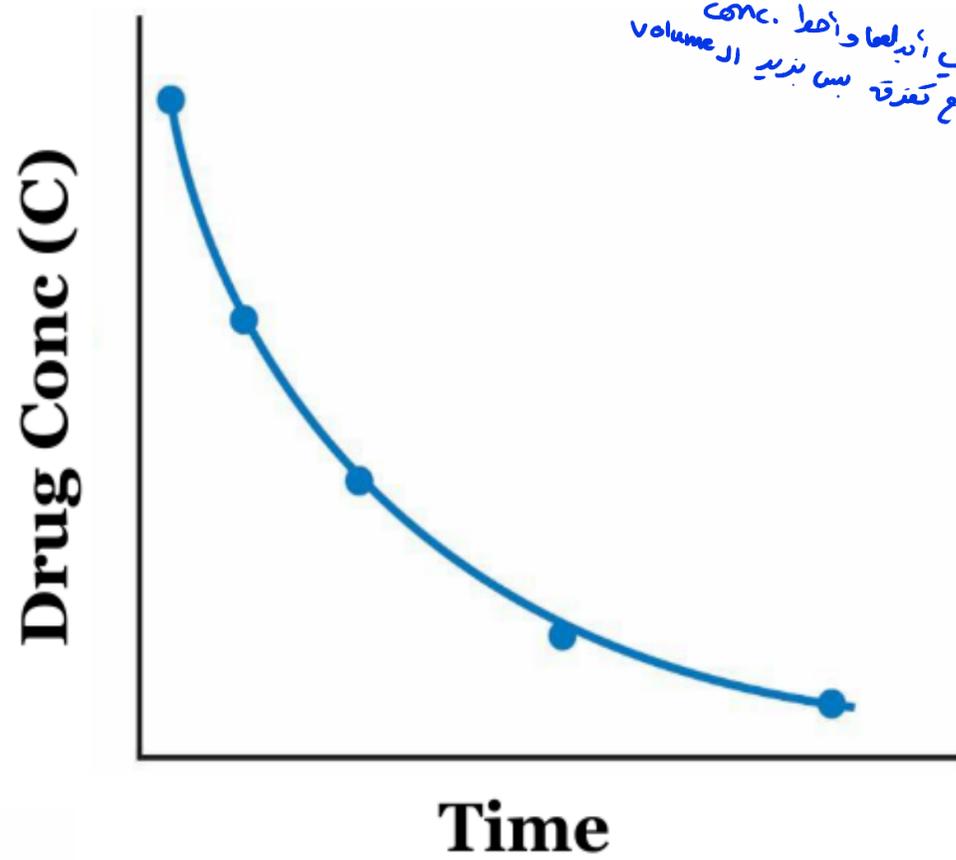
$$dD_B / dt = -k \times D_B$$

$$\ln D_B = -k \times t + \ln D_B^0 \dots \dots \dots [ln]$$

$$\log D_B = -\left(\frac{k}{2.303} \times t\right) + \log D_B^0 \dots \dots \dots [log]$$

$$D_B = D_B^0 \times e^{-kt} \dots \dots \dots [e]$$

One compartment open model



از ایدی زیاد و اصله
con. volume
ما-ع تغذیه بس بزرگ ال

$$C = \frac{D}{Vd} e^{-K \cdot t}$$

amount (drug)

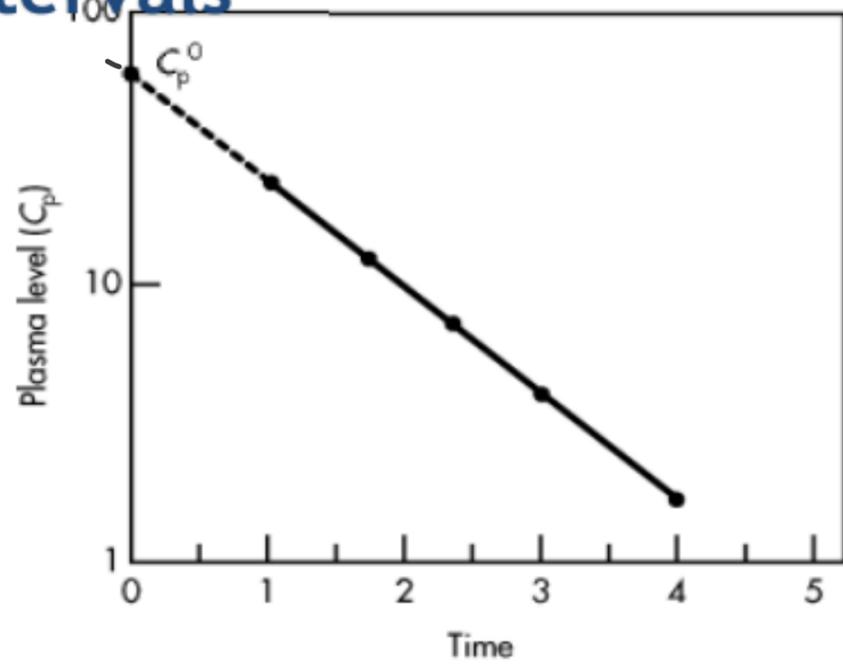
- C= concentration
- D= dose
- Vd: Volume of distribution
- K: elimination rate constant
- t: time

Calculation of concentration at different time intervals

➤ If the first plasma concentration is taken at t_1 instead of at zero and corresponds to plasma drug concentration C_1 , then C_2 is the concentration at time t_2 and t is set to $(t_2 - t_1)$ Then:

$$C_2 = C_1 * e^{-k(t_2 - t_1)} \quad \text{slope} = \frac{\Delta y}{\Delta x}$$

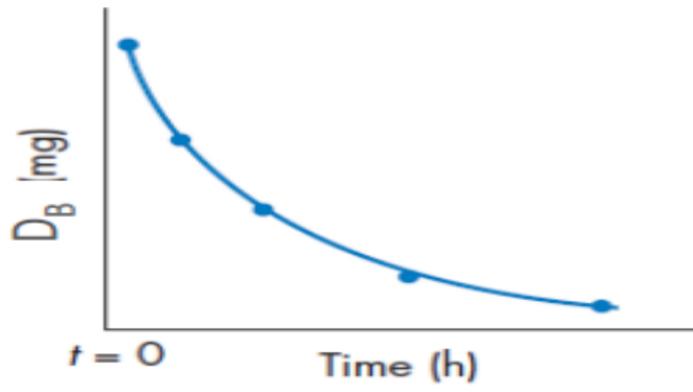
$$\ln C_2 = \ln C_1 - k(t_2 - t_1) \quad -k = \frac{\ln C_2 - \ln C_1}{(t_2 - t_1)}$$



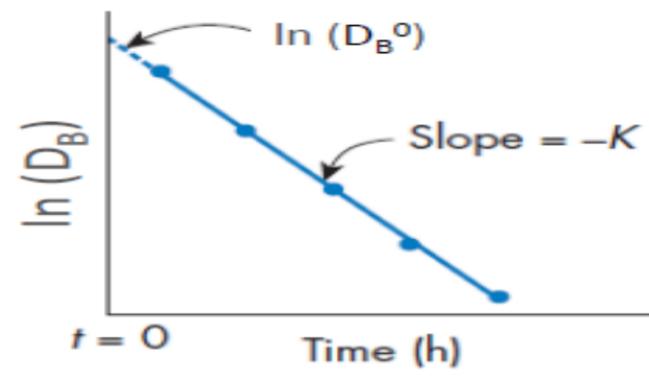
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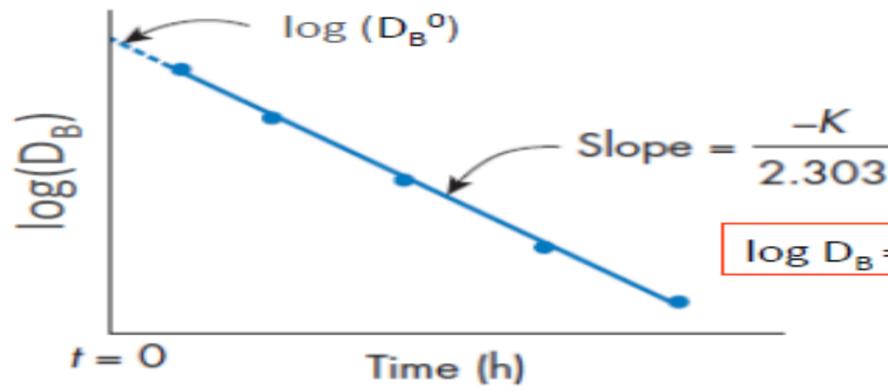
First order kinetic graphs



$$D_B = D_B^0 * e^{-kt}$$



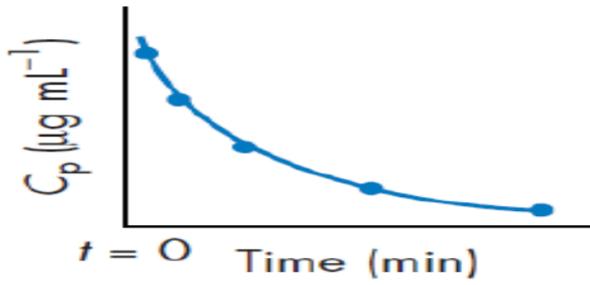
$$\ln D_B = -k*t + \ln D_B^0$$



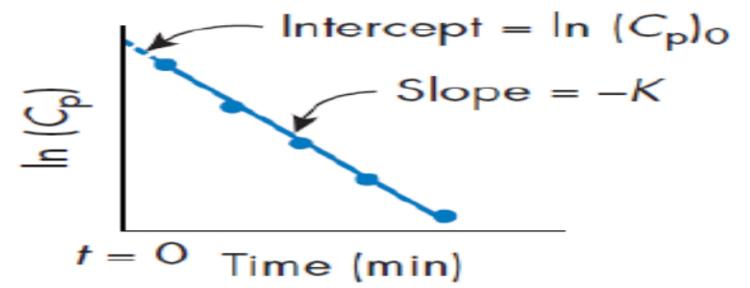
$$\log D_B = - (k*t/2.303) + \log D_B^0$$

- As we usually take samples from the plasma or blood → then the results are in concentration units not mass units
- So we will have V_D and C_p →

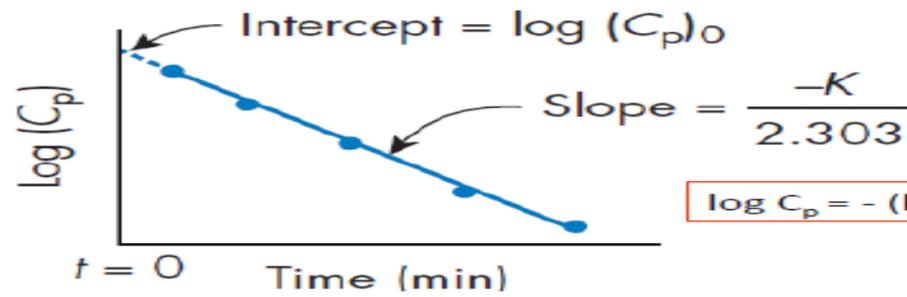
$$D_B = V_D * C_p$$



$$C_p = C_p^0 * e^{-kt}$$

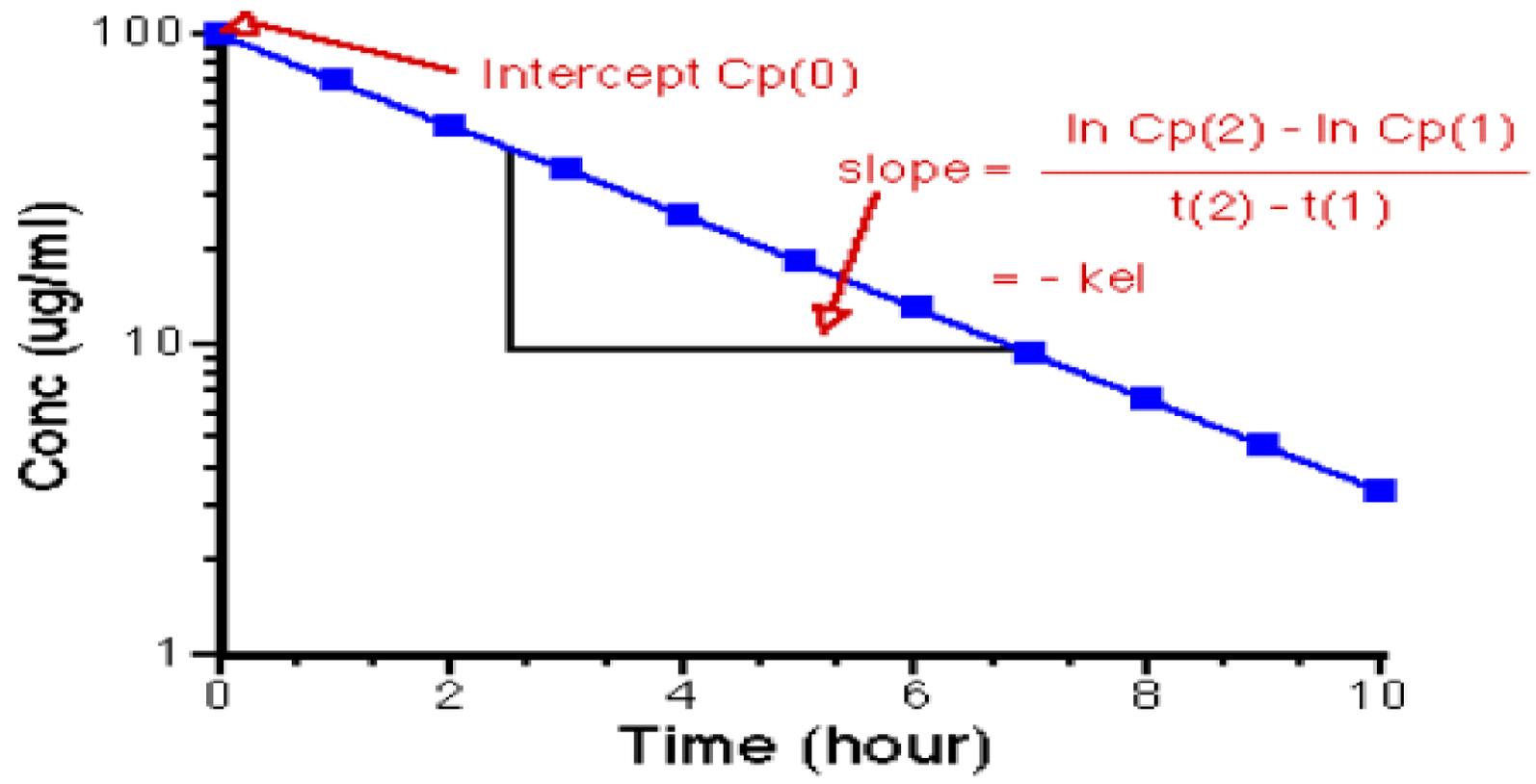


$$\ln C_p = -k*t + \ln C_p^0$$



$$\log C_p = - (k*t/2.303) + \log C_p^0$$

Determination of K



example

Practice questions

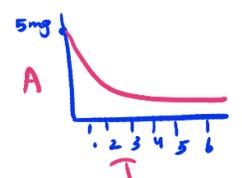
- Drug X has an elimination rate constant of 0.173 hr^{-1} , 5 mgs of the drug were administered as an IV bolus. Calculate the following

First order

$k = 0.173 \text{ hr}^{-1}$ $t = 2 \text{ hr}$
 $A_0 = 5 \text{ mg}$

- A) The drug amount after 2 hours
- B) The rate of reaction after 2 hours

① $A(t=2) = A_0 * e^{-kt}$
 $\ln A = \ln A_0 - kt$
 $\ln A = \ln 5 - 0.173 * 2$
 $\ln A = 1.6 - 0.346$
 $\ln A = 1.254$
 $A = 3.49 \text{ mg}$



$t_{1/2} = 0.693 / k$
 $= 0.693 / 0.173$

$t_{1/2} = 4$ *small hint*

③

$\frac{dA}{dt} = -kA(t=2)$
 $= -0.173 * 3.49$
 $= -0.6 \text{ mg/hr}$

عناوين اخرى rate declining
 * إذا قوت rate of reaction = -0.6 mg/hr
 * إذا قوت rate of elimination = 0.6 mg/hr
 * إذا قوت elimination rate constant = $0.173 \text{ hr}^{-1} = k$

نصفه ان بعد 4 ساعات كمية ال amount نزل لنصف من 5mg إلى 2.5mg
 فبما ان 4 نلها من 5mg بعد 2hr أي ربع كمية ال amount من 5mg إلى 2.5mg

PK parameters

Fundamental parameters in one compartment

non fundamental parameter → AUC

- Apparent Volume of Distribution (V_d)
- Elimination rate constant (K)
- Elimination half life ($t_{1/2}$)
- Clearance (Cl)

Apparent Volume of Distribution (Vd)

(ظاہری)

not real

it's not a body property, but drug property

پلاسما میں دوائی کی مقدار اور دوائی کی کل مقدار کا تناسب
plasma concentration = zero

اگر دوائی کا تمام جسم سے خارج ہو جائے تو اس کی پلاسما میں دوائی کی مقدار صفر ہوگی۔

- This apparent volume of distribution is not a physiological volume. It won't be lower than blood or plasma volume but it can be much larger than body volume for some drugs.

• It is a mathematical factor relating the amount of drug in the body and the concentration of drug in the measured compartment, usually plasma:

$$V_d = \frac{\text{AMOUNT of drug in the body}}{\text{CONCENTRATION in plasma}}$$

- Vd: A measure of the tendency of a drug to move out of the blood plasma to some other site.

Cont,

- Concentrations (mass per unit volume or amount per unit volume), not masses (mg or μg), are usually measured in plasma or serum (more often than blood).
- Therefore, a term is needed to relate the measured concentration (C_p) at a time to the mass of drug (X) at that time. This term is defined as the apparent volume of distribution (V).
- The apparent volume of distribution (V) is simply a proportionality constant whose sole purpose is to relate the plasma concentration (C_p) and the mass of drug (X) in the body at a time. **It is not a physiological volume**

$$V_d = \frac{\text{dose}}{\text{initial conc.}} = \frac{X_0}{C_0}$$

Factors Affecting Drug Distribution:

➤ Rate of distribution

- ✓ Membrane permeability → \uparrow \Rightarrow $\uparrow V_d$ (كثافة زادت \Rightarrow $\uparrow V_d$)
- ✓ Lipid Solubility \Rightarrow (partition coefficient \uparrow) \Rightarrow $\uparrow V_d$
- ✓ pH - pKa (pH-partition theory for ionizable molecules) \rightarrow \uparrow unionized \Rightarrow $\uparrow V_d$
- ✓ Blood perfusion of organs and tissues \rightarrow \uparrow \rightarrow $\uparrow V_d$

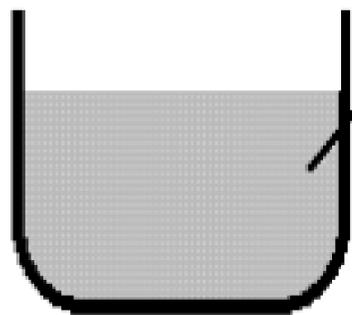
➤ Extent of Distribution

- ✓ Plasma protein binding \rightarrow \uparrow \Rightarrow $\downarrow V_d$ (مضاد البروتين \Rightarrow $\downarrow V_d$)
Free \Rightarrow $\downarrow V_d$ (Free \Rightarrow $\downarrow V_d$)
- ✓ Intracellular binding \rightarrow $\downarrow V_d$ (ينتقل ببطء في الدم \Rightarrow $\downarrow V_d$)

Volume of distribution

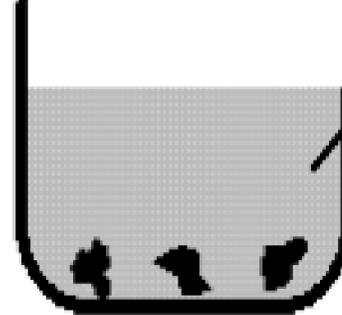
Definition: $V = \frac{\text{amount of drug in the body}}{\text{concentration measured in plasma}}$

Drug concentration in beaker:



Dose = 10 mg
 $C_p^0 = 20 \text{ mg/L}$
Apparent
Volume = 500 ml

With charcoal in beaker:



Dose = 10 mg
 $C_p^0 = 2 \text{ mg/L}$
Apparent
Volume = 5000 ml

Cont,

The more the drug penetrate into tissues/organs following the administration of the dose, the smaller will be the plasma and/or serum drug concentration → →

Therefore the higher is the hypothetical volume into which the drug is distributed

- **V_D is usually a property of a drug rather than of a biological system:** the extent to which certain drug is distributed in the body tissues

Cont,

- Reflects the extend of drug distribution in the body tissues and organs

↑ drug distribution → ↑ V_D

e.g.

- Highly protein bound or highly water soluble drugs

→ ↓ distribution → ↓ V_D

- Drugs accumulated in adipose tissues → ↑ V_D

- Reflects the lipophilicity of a drug

↑ drug lipophilicity → ↑ V_D

↑ drug hydrophilicity → ↓ V_D

List of volume of distribution of some drugs

Volume of Distribution

(volume of distribution coefficient) * صافي وصة (معدل ولا جا)

صافي مش وصة volume ← وصة ل volume (L) أو (ml) كالمنا بعضه اشترى جانبها

Erythropoietin	5 L	0.07 L/kg*
Warfarin	8 L	0.12 L/kg*
Phenytoin	45 L	0.63 L/kg*
Digoxin	500 L	7 L/kg*
Amiodarone	5000 L	70 L/kg*
Chloroquine	15000 L	215 L/kg*
Quinacrine	35000 L	500 L/kg*

* ليع 70 kg ؟
الوضع الطبيعي لما
نجم نسبة وصة الانسان
بشروط انه وصة
70 kg
مثلا :
500 L x 70 kg
kg
= 35000 L

antimabials

disposition ← نقيضه صافي وصة ← كموثلية صافي الير يصل موجود بالدم ، الياح كيه صافي disposition

* Distribution Coefficient

Clearance ← $Cl = k \cdot Vd$

Cl ?

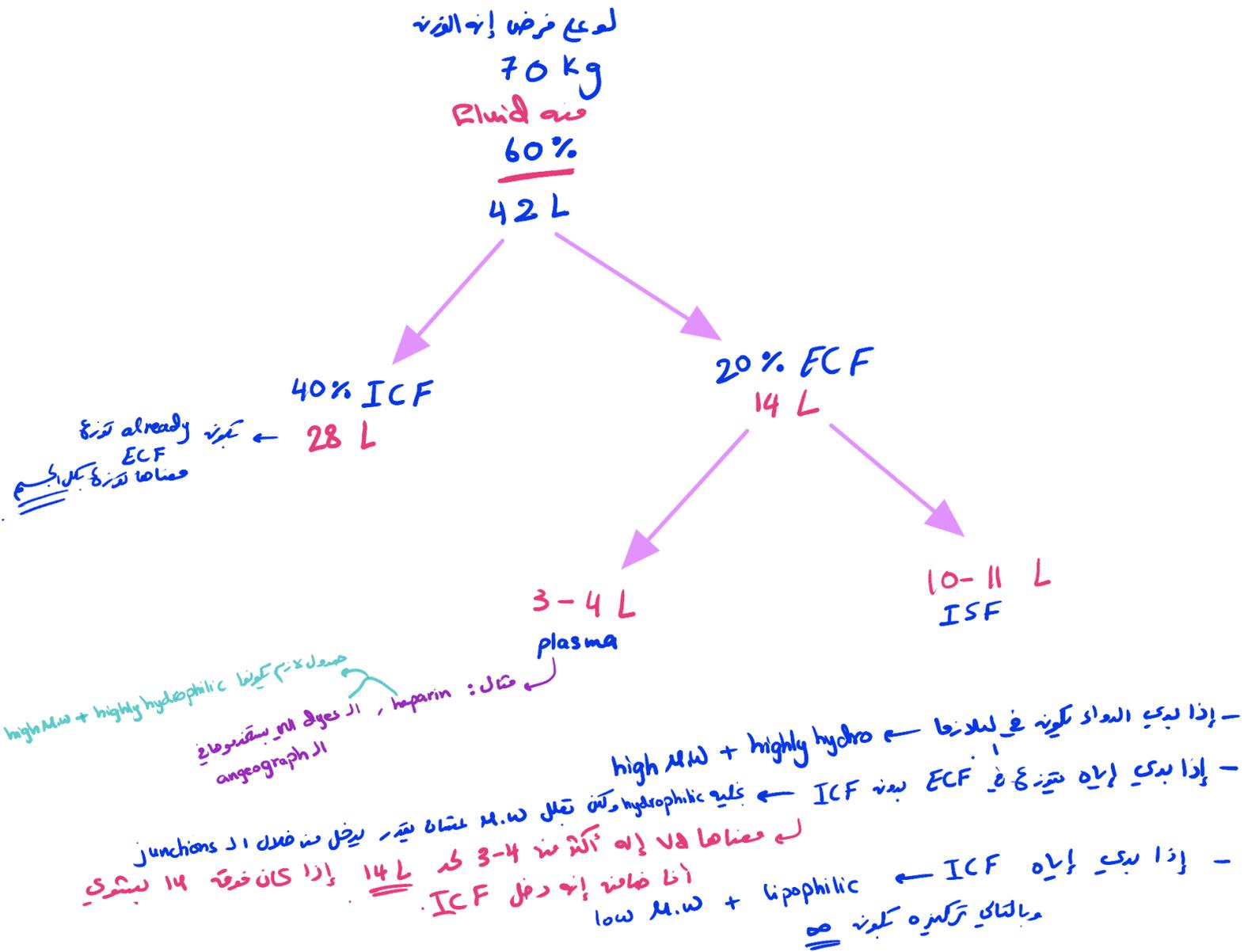
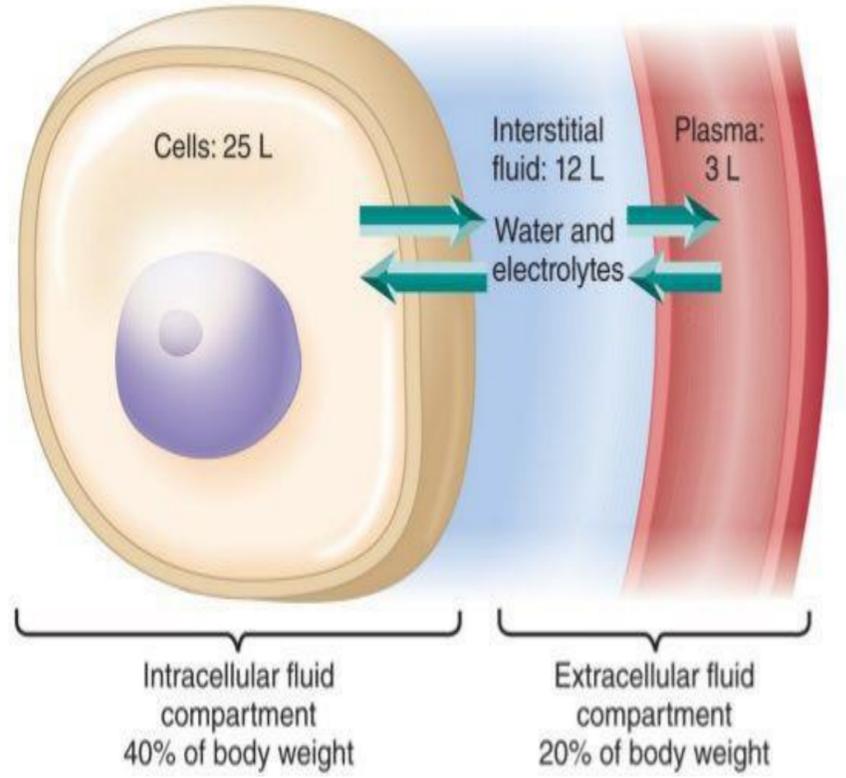
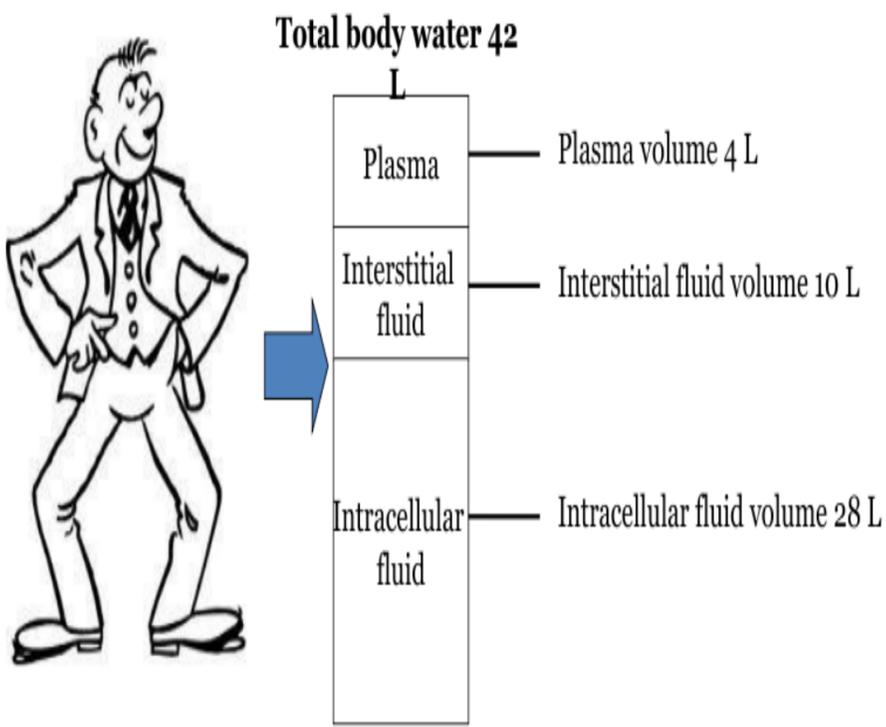
$V = 2 \text{ L/kg} \rightarrow 2 \times 70 = 140 \text{ L}$

$k = 0.1$

$Cl = 0.1 * 140 = 14 \text{ L/hr}$

The real Volume of Distribution has physiological meaning and is related to body water

Major fluid compartments in the body



Apparent Volume of Distribution

- If a drug has a high molecular weight or is extensively protein bound, it is too large to pass through the slit junctions of the capillaries and, thus, is effectively trapped within the plasma (vascular) compartment. As a result, it has a low V_d that approximates the plasma volume, or **about 4 L in a 70-kg individual** (e.g. Heparin).
- If a drug has a low molecular weight but is hydrophilic, it can pass through the endothelial slit junctions of the capillaries into the interstitial fluid. However, hydrophilic drugs cannot move across the lipid membranes of cells to enter the intracellular fluid. Therefore, these drugs distribute into a volume that is the sum of the **plasma volume and the interstitial fluid**, which together constitute the extracellular fluid, (about 20% of body weight or 14 L in a 70-kg individual) (e.g. aminoglycoside antibiotics)

Apparent Volume of Distribution

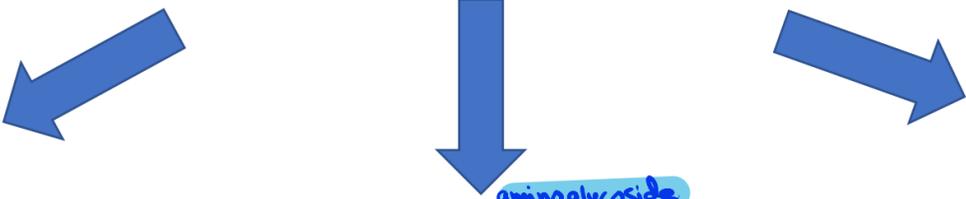
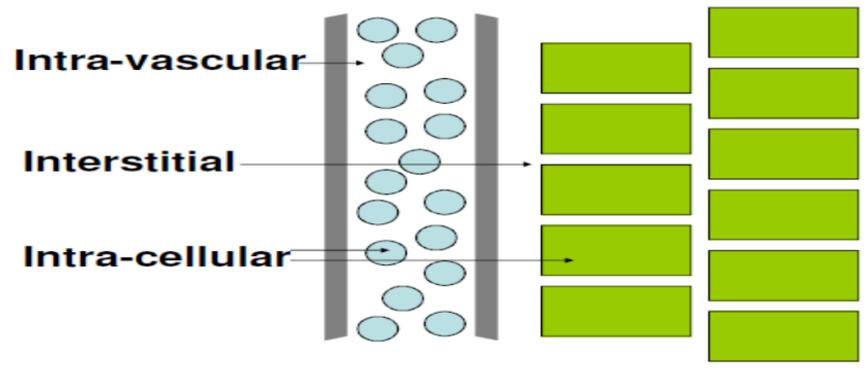
- If a drug has a low molecular weight and has enough lipophilicity, it can move into the interstitium through the slit junctions and pass through the cell membranes into the intracellular fluid. These drugs distribute into a volume of about 60% of body weight or about 42 L in a 70-kg individual. *Ethanol* exhibits this apparent Vd.
- In general, a larger Vd indicates greater distribution into tissues; a smaller Vd suggests confinement to plasma or extracellular fluid.

Drug X has a volume of distribution of 20 ^{Vd coefficient} (L/kg), what does this mean?

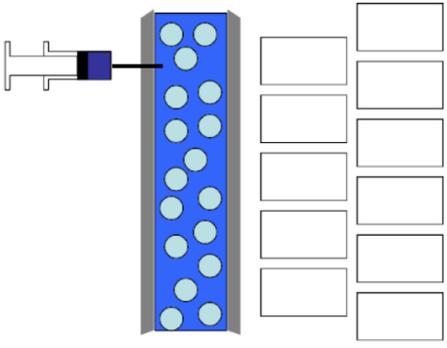
Volume of distribution

- The apparent volume of distribution has a minimum value that is **dependent on physiological factors**. A drug must be distributed at least throughout the plasma. Therefore, **the minimum** value of the apparent volume of distribution should be at least 3–4 L in a healthy 70 kg subject.
- There is **theoretically**, however, **no upper limit**. The higher the tissue affinity, the lower the fraction of drug will be in plasma.
- Theoretically, if the plasma concentration approaches a **value of zero at infinitely high tissue affinities**, **the value of the volume of distribution moves towards infinity**. ∞

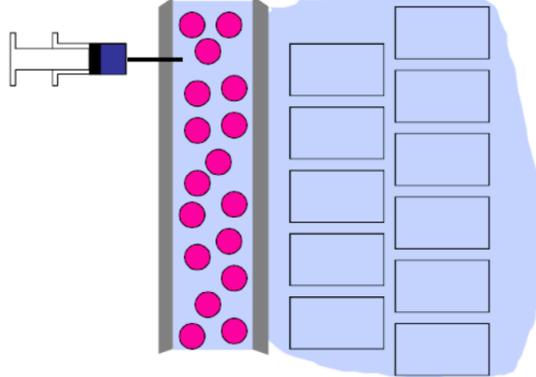
Body water



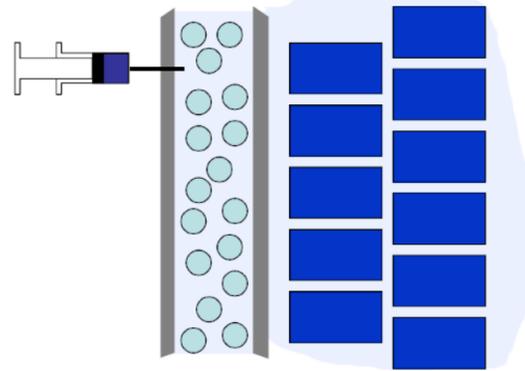
Distribution - Evan's Blue
Intra-vascular space only



Distribution - ~~Ethanol~~ aminoglycoside
All water



Distribution - Quinacrine
Concentration into cells



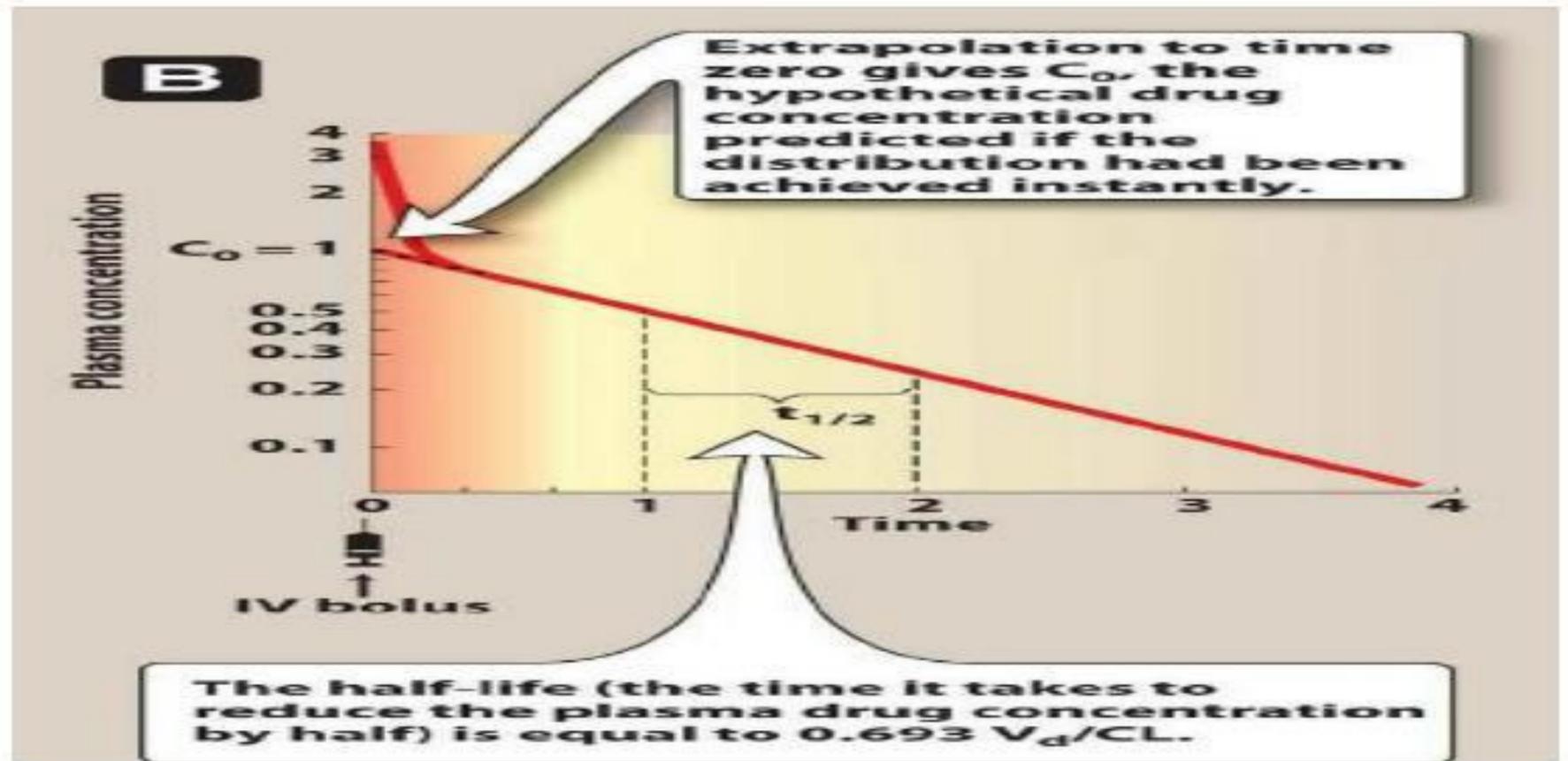
Apparent volume of distribution estimation

1. Plot $\log(C)$ vs. time
2. Plot the best-fit line
3. Extrapolate to the Y-axis intercept (to estimate initial concentration, C_0)

4. Estimate V_d :
$$V_d = \frac{\text{dose}}{\text{initial conc.}} = \frac{X_0}{C_0}$$

vol. = $\frac{\text{amount}}{\text{conc.}}$

Apparent Volume of Distribution: Mathematics



Cont,

- In a one-compartment model (IV administration), the V_D is calculated with the following equation:

$$V_D = \frac{\text{Dose}}{C_p^0} = \frac{D_B^0}{C_p^0}$$

- When C_p^0 is determined by extrapolation, C_p^0 represents the instantaneous drug concentration after drug equilibration in the body at $t = 0$.
- The dose of drug given by IV bolus (rapid IV injection) represents the amount of drug in the body, D_B^0 , at $t = 0$. Because both D_B^0 and C_p^0 are known at $t = 0$, then the

$$\frac{dD_B}{dt} = -kD_B$$

the second pharmacokinetic non fundamental parameter

Extracting AUC

المساحة تحت المنحنى

$$\frac{dD_B}{dt} = -kD_B$$

Substituting $D_B = V_D C_P$ into the previous Equation, the following expression is obtained: - جوانبها AUC :

$$\frac{dD_B}{dt} = -kV_D C_P$$

1

$$AUC = \frac{D_0}{k \cdot V_d}$$

$$dD_B = -kV_D C_P dt$$

2

$$AUC = \frac{C}{k}$$

$$\int_0^{D_0} dD_B = -kV_D \int_0^{\infty} C_P dt$$

Volume of distribution vs AUC

The integral $\int_0^{\infty} C_p dt$ represents the AUC_0^{∞} , which is the summation of the area under the curve from $t = 0$ to $t = \infty$. Thus, the apparent V_D may also be calculated from knowledge of the dose, elimination rate constant, and the area under the curve (AUC) from $t = 0$ to $t = \infty$. This is usually estimated by the trapezoidal rule (see Chapter 2). After integration, Equation 4.12 becomes

$$D_0 = kV_D [AUC]_0^{\infty}$$

which upon rearrangement yields the following equation:

$$V_D = \frac{D_0}{k [AUC]_0^{\infty}} \quad (4.13)$$

**Failure is the key
to SUCCESS.
Each mistake
teaches us something.**

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