

تفريغ حساب وتركيب الأشكال الصيدلانية



اسر الموضوع: Suspensions ▼

إعداد الصيدلاني /^ة: ياسمين خليل ▼



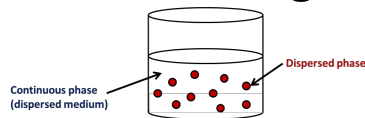
لجان الرِّفعات

﴿ وَمَا رَمَيْتَ إِذْ رَمَيْتَ وَلَكِنَّ اللَّهَ رَمَى ﴾



Part 1: Introduction and advantages

Definition



- A suspension consists of a dispersion of insoluble solid particles in a liquid.

- In a suspension, the particle size is generally > 1 μm. مفوض
- Different to a colloidal system, where particle size is < 1 μm.

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Solid in liquid colloids vs. solutions micible

- Suppose we add a powder of drug to water. How do I know if I have a colloid or a solution?
- In a true solution, the drug molecules are dispersed on the molecular scale in the solution – each molecule is isolated from all other drug molecules;
- In a colloid, we have particles of drug – aggregates زیر کتل consisting of many drug molecules.

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Stokes' law

σ : density of solid
علاقة طردية مع sedimentation
 ρ : density of liquid
علاقة عكسية مع sedimentation

Given that:

$$v = \frac{2a^2 g (\sigma - \rho)}{9\eta}$$

طردية [σ, a]
عكسية [ρ, η]

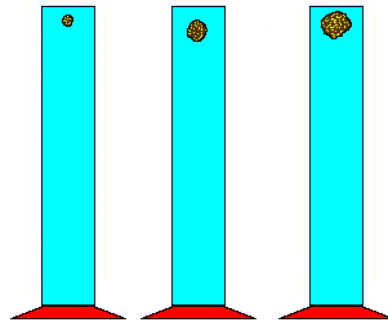
Will the sedimentation velocity go up or down if we:

- Increase the radii of the particles? \uparrow
- Increase the density of the solid? \uparrow
- Increase the density of the liquid? \downarrow
- Decrease the density of the liquid? \uparrow
- Increase the viscosity of the liquid? \downarrow
- Decrease the viscosity of the liquid? \uparrow

Stokes' law

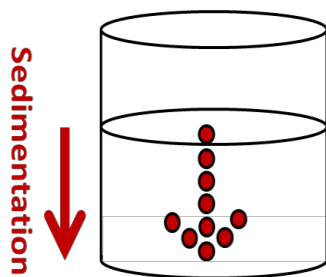
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- Increase the density of the liquid?
- Decrease the density of the liquid?
- Increase the viscosity of the liquid?
- Decrease the viscosity of the liquid?



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Sedimentation and suspension



- The velocity of sedimentation is given by Stokes' Law:

$$v = \frac{2a^2 g(\sigma - \rho)}{9\eta}$$

a = the radius of the solid particles; *طردی*

σ = the density of the solid; *طردی*

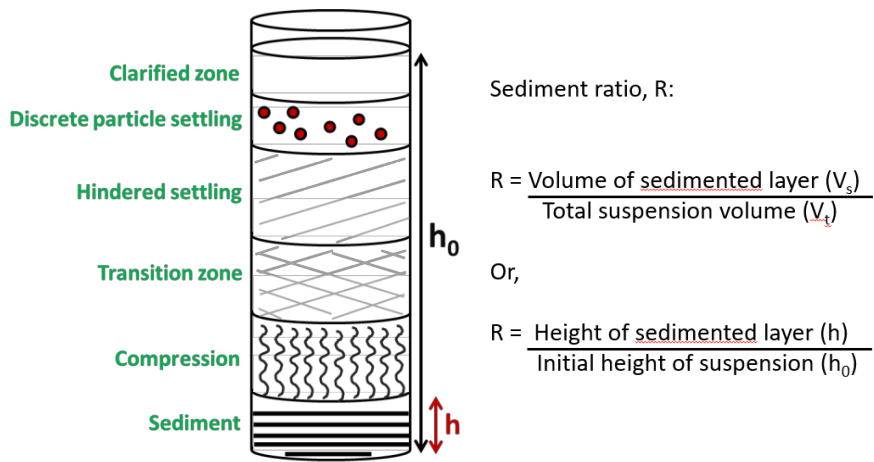
ρ = the density of the liquid; *عکسی*

η = the viscosity of the liquid; *عکسی*

g = the acceleration due to gravity. *ثابت = 9.8 m/s*

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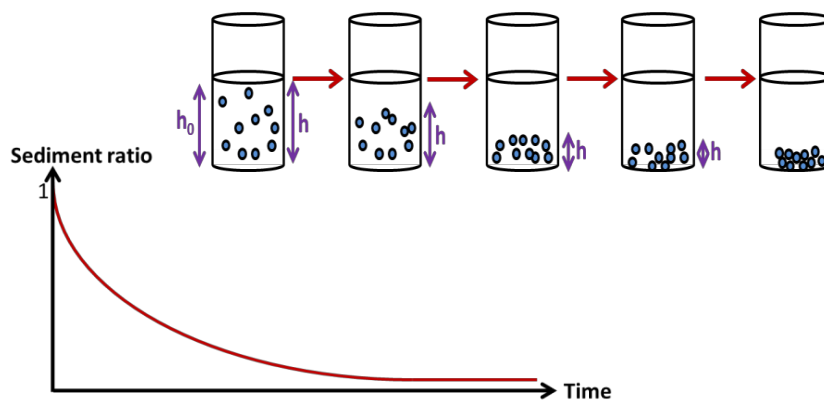
The sediment ratio



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The sediment ratio

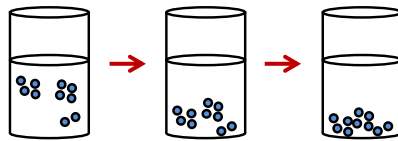
$$R = \frac{\text{Height of sedimented layer } (h)}{\text{Initial height of suspension } (h_0)}$$



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Flocculation

- In a **flocculated system**, the aggregates settle quickly. This leads to liquid entrapment in the sediment, which tends to be fairly easy to redisperse.



- So, in pharmaceuticals, **flocculated suspensions are better than deflocculated ones** (why do you think this is?)!

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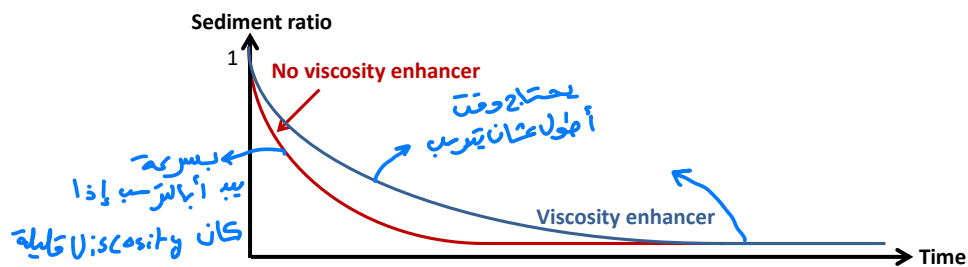
Flocculated vs. deflocculated

Flocculated system	Deflocculated systems
Loose <u>aggregates</u> of particles	Particles exist as discrete units
<u>Large volume of final sediment</u>	<u>Small volume of final sediment</u>
<u>Rapid sedimentation rate</u>	<u>Slow sedimentation rate</u>
<u>Suspension clears quickly</u>	Suspension remains cloudy for a <u>prolonged period of time</u>
Entrapment of liquid within <u>suspension</u> <u>سحب سائل من تعليق</u>	Liquid entrapment in the sediment is prevented
<u>Easy to redisperse sediment</u>	<u>Difficult to redisperse sediment</u>

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Viscosity enhancing agents

$$R = \frac{\text{Height of sedimented layer (h)}}{\text{Initial height of suspension (h}_0\text{)}}$$



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Viscosity enhancing agents

Examples of viscosity enhancing agents include:

- Polysaccharides (acacia, alginates, tragacanth, starch, xanthum gum).
الاعتماد شوي
- Celluloses (methylcellulose, hydroxyethylcellulose, sodium carboxymethylcellulose).
- Hydrated silicates (bentonite, magnesium aluminium silicate).
- Carbomers and silicon dioxide (Aerosil).

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